Coupled Cluster Calculations of the Ground and Excited Electronic States Using Two- and Four-Component Relativistic Spinors

Rajat K. Chaudhuri

Indian Institute of Astrophysics, Bangalore-560034, India. E-mail: rajat@iiap.ernet.in

Received: 7 September 2003 / Accepted: 28 October 2003 / Published: 4 December 2003

Abstract: The coupled cluster based linear response theory which is applicable to the direct calculation of atomic and molecular properties are presented and applied to compute the ionization potentials and excitation energies of light and moderately heavy atoms. The effect of electron correlation on the ground and excited states is studied using Hartree-Fock, Dirac-Fock and approximate two-component relativistic spinors.

Keywords: Standard Model, Model space, Kinetic balance, Variational collapse, HF, DF, RESC, CCM and CCLRT.

©2003 by MDPI, Basel, Switzerland. Reproduction for noncommercial purposes permitted.

1 Introduction

The study of physical and chemical processes in atomic and molecular systems containing heavy elements is of great importance both experimentally and theoretically. Aside from their obvious relevance to understanding the chemistry of compounds containing heavy atoms, these studies can provide a quantitative estimate of relativistic contributions, kinematic effects, and a probe for physics that departs from the predictions of the *standard model*. The twin facts that heavy atom compounds contain many electrons and that the behavior of these electrons must be treated relativistically introduce severe impediments to accurate theoretical treatments of systems containing heavy atoms. Rigorous relativistic electronic structure methods begin by forgoing the Schrödinger equation in favor of the Dirac equation. The concomitant size of the matrices to be manipulated and the number of two-electron integrals to be evaluated in the Dirac equation have forced the introduction of various approximate method to describe the electronic structure of heavy atom polyatomics by either restricting the number of electrons to be treated explicitly or by converting the relativistic problem into a combination of a non-relativistic many electron problem, a perturbative treatment of the relativistic corrections, and/or both.

The most widely used approximate relativistic scheme for describing heavy atom systems is the *effective core potential* (ECP) method, where the core electrons are represented by suitable functions and where only the valence electrons are treated explicitly. Recently Motegi *et al.* have proposed a somewhat less approximate relativistic scheme (RESC) [1] for generating relativistic spinors. The RESC method proceeds by eliminating the small component portion of the relativistic Hamiltonian from the four-component Dirac equation through a suitable transformation.

In this article, we illustrate the efficacy of the RESC scheme by comparing the ionization potentials (IPs) and excitation energies (EEs) obtained from the coupled cluster based linear response theory (CC-LRT) [2–10] with Hartree-Fock, RESC, and four-component Dirac-Fock orbitals.

2 Coupled Cluster Based Linear Response Theory

In CC-LRT approach, one begins with the ground state function $|\Psi_0\rangle$, written in the coupled cluster form

$$|\Psi_0\rangle = exp(T)|\phi_0\rangle \tag{1}$$

where the cluster operator T consists of various hole-particle excitations from the closed-shell function ϕ_0 , taken as the vacuum. The ionized/excited state Ψ_k are generated from the ground state by the action of ionization/ excitation operator W_k^{\dagger} , and the corresponding energy ω_k are obtained from an equation of the form [2–7]:

$$[H, W_k^{\dagger}]|\Psi_0\rangle = \omega_k W_k^{\dagger}|\Psi_0\rangle \tag{2}$$

In non-orthogonally spin-adapted formulation of CC-LRT, both T and W_k^{\dagger} are expressed as a linear combination of various spin-free hole-particle excitations operators, written in normal-order. Thus, for example, for single and double excitations, one has operators like

$$q_{p\alpha}^{\dagger} = \sum_{\sigma} a_{p\sigma}^{\dagger} a_{\alpha\sigma} = \{e_{p\alpha}\},\tag{3}$$

and

$$q_{p\alpha,q\beta}^{\dagger} = \{e_{p\alpha}, e_{q\beta}\},\tag{4}$$

respectively.

The curly brackets in eqs. (3) and (4) stand for normal ordering. The labels α , β and p, q signify hole and particle orbitals, respectively, and σ denotes spin labels. The operators T and W_k^{\dagger} are expressed as

$$T = \sum_{i} t_i q_i^{\dagger},\tag{5}$$

and

$$W_k^{\dagger} = \sum_i x_{ik} q_i^{\dagger}, \tag{6}$$

respectively, for excitation process. The cluster operators $W_k^{\dagger}s$ for ionization and electron attachment processes are expressed as

$$W_k^{\dagger} = \sum_{\alpha=1}^{occ} x_{\alpha} a_{\alpha} + \sum_{\alpha,\beta=1}^{occ} \sum_{p=1}^{unocc} x_{\alpha\beta}^p a_{\beta}^{\dagger} a_{\beta} a_{\alpha} \quad \text{(for IP)}, \tag{7}$$

and

$$W_k^{\dagger} = \sum_{p=1}^{unocc} x_p a_p^{\dagger} + \sum_{p,q=1}^{uocc} \sum_{\alpha=1}^{occ} x_{\alpha}^{pq} a_p^{\dagger} a_q^{\dagger} a_{\alpha} \quad \text{(for EA)}, \tag{8}$$

respectively.

Since T and W_k^{\dagger} commute, premultiplying eq. (2) by exp(-T) we get the following equation of motion (EOM):

$$[\bar{H}, W_k^{\dagger}] |\phi_0\rangle = \omega_k W_k^{\dagger} |\phi_0\rangle \tag{9}$$

with

$$\bar{H} = exp(-T)Hexp(T) + E_0 \tag{10}$$

where E_0 is the ground state energy. Projecting eq.(9) on to the biorthogonal space $\langle \tilde{\phi}_i | = \langle \phi_0 | \tilde{q}_i,$ we get

$$AX_k = \omega_k X_k,\tag{11}$$

where

$$A_{ij} = \langle \phi_0 | \tilde{q}_i [\bar{H}, q_j^{\dagger}] | \phi_0 \rangle \tag{12}$$

and X_k denotes the columns of x_k . Since T is anti-Hermitian, eq. (10) leads to a non-Hermitian operator \overline{H} and the associated matrix A.

3 Results

In the present work, we have used the truncation scheme $T = T_1 + T_2$ for computing Ψ_0 . The excited/ionized states Ψ_k are generated by the action of W_k^{\dagger} where the excitation/ionization operator W_k^{\dagger} is truncated at 2h-2p level for EE, 2h-1p for IP, and 1h-2p for EA processes, respectively. To avoid variational collapse in the DF procedure, we impose kinetic balance condition and relativistic boundary conditions on the basis spinors. Since the expression for kinetic balance condition is simple for uncontracted Gaussian/Slater functions, we use uncontracted Gaussian basis to generate the Hartree-Fock(HF), RESC and Dirac-Fock (DF)orbitals. While the HF and RESC orbitals are obtained from GAMESS [11] code, the DF orbitals are generated from the code developed in our laboratory. Note that in the present RESC scheme, only the one electron integral corrections are added. A more accurate treatment can be achieved by means of relativistically averaged basis set which incorporates the two- electron integral corrections. However, the RESC corrections to two-electron is beyond the scope of the present work.

The basis sets used in the computation of IPs and EEs are displayed in Table 1. Figure 1 plots the absolute difference in the ground state energy obtained from CC calculation using DF, HF, and RESC orbitals with respect to atomic number (Z), i.e., $|E_{gr}^{DF} - E_{gr}^{HF}|$ and $|E_{gr}^{DF} - E_{gr}^{RESC}|$ versus the atomic number. The present work shows that the ground energy obtained from HF orbital is less than that obtained from the RESC and DF orbitals. Figure I also demonstrates that while $|E_{gr}^{DF} - E_{gr}^{HF}|$ increases with increasing atomic number, E_{gr}^{RESC} remains quite close to E_{gr}^{DF} over the entire range. The fact that $|E_{gr}^{DF} - E_{gr}^{RESC}| << |E_{gr}^{DF} - E_{gr}^{HF}|$ for the ground state clearly demonstrates that dominant relativistic corrections can be incorporated through RESC scheme.

Table 2 compares the ionization potentials (IPs) obtained from CCLRT-IP method using HF, DF, and RESC orbitals with the experiment [12]. Table 2 shows that the average deviation (absolute) of the computed IPs with HF, RESC and DF orbitals are 1.269%, 0.576% and 0.552%, respectively. Table II further demonstrates that the accuracy of computed IPs with HF orbitals

System	Basis set	Ref.
Be	16s9p5d1f	[15]
В	16s10p5d1f	[15]
С	14s10p4d1f	[15]
Ne	13s9p5d2f	[16]
Ar	20s13p4d2f	[17]
Rb	15s12p8d2f	[18]
\mathbf{Cs}	15s12p8d2f	[18]

Table 1: Basis set used in the computation IPs and EEs.



Figure 1: Plot of $|E_{gr}^{DF} - E_{gr}^{HF}|$ (solid line) and $|E_{gr}^{DF} - E_{gr}^{RESC}|$ (dotted line) versus the atomic number (Z).

Int. J. Mol. Sci. 2003, 4

System		CCLRT		Experiment [12]
	HF	RESC	DF	
Be	9.315	9.316	9.322	9.322
В	8.222	8.220	8.223	8.298
B^+	25.134	25.140	25.146	25.155
C^+	24.292	24.289	24.289	24.383
Ne	21.240	21.262	21.633	21.564
Ar	15.656	15.638	15.898	15.759
Rb	4.074	4.141	4.185	4.177
Cs	3.735	3.889	3.960	3.894
Average Error	1.269%	0.576%	0.552%	

Table 2: Ionization Potentials (in eV) of Be, B, C⁺⁺, Ne, Ar, Rb, and Cs obtained from CCLRT-IP method using Hartree-Fock (HF), RESC, and Dirac-Fock (DF) orbitals.

deteriorates with increasing atomic number (Z) as anticipated. The fact that IPs obtained using RESC orbitals are quite close to those obtained from DF orbitals suggests that relativistic effects can be introduced efficiently via RESC scheme.

Table 3 compares the CCLRT valence electron removal energies of B, C⁺, Rb, and Cs atom with the experiment [12]. It is evident from Table 3 that the average error in computed valence electron removal energies with HF orbitals is large compared to those obtained from RESC and DF orbitals. Table 3 also indicates that the computed valence electron removal energies of C⁺(${}^{2}P_{1/2}$), Rb(${}^{2}S_{1/2}$), and Cs(${}^{2}S_{1/2}$) states are quite off from the experiment. However, this is purely a basis inadequacy problem. Using larger basis (with more d and f) this difference can be reduced significantly [13]. Since, the main motivation of this work is to assess the relative accuracy of different choice of orbitals, we use limited but same basis for all three cases.

In Table 4, we compare the excitations energies of Be, B⁺, and C⁺⁺ obtained from CCLRT-EE method using HF and RESC orbitals with the experiment. Table IV shows that the average and RMS errors are slightly less for EE obtained from the HF orbitals than those computed with RESC orbitals. Since the systems investigated here are not heavy, it is difficult at this stage to assess the efficacies of the RESC scheme. However, based on the previous results (displayed in Tables 2 and 3), we strongly believe that the accuracy of estimated EEs from HF orbitals will deteriorate with increasing atomic number. Moreover, studies on alkali metal halides (NaCl, NaBr, KCl, RbCl etc.) suggest that the RESC orbitals offers more accurate estimate of ionization potentials, electron affinities, and dissociation energies than the HF orbitals [14] for systems containing heavy elements.

Int. J. Mol. Sci. 2003, 4

System	Dominant		CCLRT		Experiment [12]
	Configuration	HF	RESC	DF	
В	$[Be]2p(^{2}P_{3/2})$			66322	66946
	$[Be]3s(^2S_{1/2})$	26787	26790	26723	26890
	$[Be]4s(^2S_{1/2})$	11829	11830	11789	11921
C^+	$[Be]2p(^{2}P_{3/2})$			195903	196723
	$[Be]3p(^2P_{1/2})$			64551	64934
	$[Be]3p(^{2}P_{3/2})$	64786	64783	64780	64923
Rb	$[Kr]6s(^2S_{1/2})$	10963	11242	11260	13557
	$[Kr]5p(^{2}P_{1/2})$			20804	21112
	$[Kr]5p(^{2}P_{3/2})$	20481	20500	20577	20874
Cs	$[Xe]7s(^{2}S_{1/2})$	9067	9854	9960	12872
	$[Xe]6p(^{2}P_{1/2})$			19741	20229
	$[Xe]6p(^{2}P_{3/2})$	19215	19265	19309	19674
Average Error		7.754%	6.534%	6.397%	

Table 3: Valence electron removal energies (in cm^{-1}) of B, C⁺, Rb, and Cs obtained from CCLRT-EA method using Hartree-Fock (HF), RESC, and Dirac-Fock (DF) orbitals.

4 Discussion

The coupled cluster based linear response theory is presented and applied to compute ionization potentials and excitation energies for light atoms. In this work, we investigate the effect of electron correlation on the ground and excited states using HF, DF and RESC orbitals. The present work demonstrates that dominant relativistic corrections can be incorporated through RESC scheme. The present work further demonstrates that for heavy elements accuracy of the properties computed with HF orbitals is much less than those obtained from DF and RESC orbitals. Since the two-electron integral corrections are not included in the present RESC scheme, comparison between the DF and RESC scheme is not perfect at this stage. A comparative study of DF orbitals versus RESC with relativistically averaged basis set which incorporates the twoelectron integral corrections will be more appropriate. Research in this direction is in progress.

Int. J. Mol. Sci. 2003, 4

System	Dominant	CCLRT		Experiment [12]
	Configuration	HF	RESC	
Be	$[{\rm He}]2s3s({}^{1}S)$	54657	54665	54662
	$[\mathrm{He}]2s2p(^{1}P)$	42806	42819	42565
	$[\mathrm{He}]2s3p(^{1}P)$	60418	60429	60187
	$[{\rm He}]2s4s({}^{1}S)$	65222	65230	65245
	$[\mathrm{He}]2s5s(^{1}S)$	70107	70115	69332
B+	[He] $2s2p(^{1}P)$	73993	74052	73397
	[He] $2p^2(^1D)$	104338	104449	102362
	[He] $2p^2({}^1S)$	129283	129368	127622
	[He] $2s3s(^{1}S)$	138333	138396	137622
	[He] $2s3p(^{1}P)$	144079	144131	144103
	[He] $2p^2(^1D)$	156856	156907	154687
	[He] $2s4s(^{1}S)$	167801	167845	167936
	[He] $2s4p(^{1}P)$	171322	171374	170591
C++	[He] $2s2p(^{1}P)$	103254	103411	102351
	[He] $2p^2(^1D)$	148668	148966	145875
	[He] $2s3s(^{1}S)$	185593	185883	182520
Average Error		0.82%	0.87%	

Table 4: Singlet state excitation energies (in cm^{-1}) of Be, B⁺ and C⁺⁺ obtained from CCLRT-EE using Hartree-Fock (HF) and RESC orbitals.

References

- Nakajima, T.; Suzumura, T.; Hirao, K. Chem. Phys. Letts. 1999, 304, 271; Suzumura, T.; Nakajima, T.; Hirao, K. Int. J. Quantum Chem. 1999, 75, 757; Nakajima, T.; Koga, K.; Hirao, H. J. Chem. Phys. 2001, 112, 10142; Motegi, K.; Nakajima, T.; Hirao, K.; Seijo, L. J. Chem. Phys. 2001, 114, 6000.
- 2. Mukherjee, D. and Mukherjee, P. K. Chem. Phys. 1979, 37, 327.
- 3. Ghosh, S.; Bhattacharyya, S. N.; Mukherjee, D. Chem. Phys. 1982, 72, 961.
- 4. Adnan, S. S. Z.; Bhattacharyya, S. N.; Mukherjee, D. Mol. Phys. 1982, 39, 519.

- 5. Adnan, S. S. Z.; Bhattacharyya, S. N.; Mukherjee, D. Chem. Phys. Letts. 1982, 85, 204.
- Mukhopadhayay, S.; Chaudhuri, R. K.; Mukhopadhyay, D.; Mukherjee, D. Chem. Phys. Letts. 1990, 173, 181.
- 7. Kundu, B. and Mukherjee, D. Chem. Phys. Letts. 1991, 179, 468.
- Monkhorst, H. J. Int. J. Quantum Chem. 1977, S11, 421; Dalgaard, E. and Monkhorst, H. J. Phys. Rev. 1983, A 28, 1217.
- Sekino, H. and Bartlett, R. J. Int. J. Quantum Chem. 1984, S 18, 255; Greetsen, J.; Rittby, M.; Bartlett, R. J. Chem. Phys. Lett. 1989, 164, 57.
- Koch, H.; Jensen, H. J. A.; Jorgensen, P.; Helgaker, T. J. Chem. Phys. 1990, 93, 3345; Koch, H. and Jorgensen, P. J. Chem. Phys. 1990, 93, 3333.
- Schmidt, M.W.; Baldridge, K.K.; Boatz, J.A.; Elbert, S.T.; Gordon, M.S.; Jensen, J.H.; Koseki, S.; Matsunaga, N.; Nguyen, K.A.; Su, S.J.; Windus, T.L.; Dupuis, M.; Montgomery, J.A. J.Comput.Chem. 1993, 14, 1347.
- Moore, C. E. Atomic Energy Levels, Natl. Bur. Standard, Ref. Data Ser., Natl. Bur. Stand. (U.S.) Circ. No. 35 (U.S. GPO, Washington, D. C., 1971), Vol 1.
- Chaudhuri, R. K.; Sahoo, B. K.; Das, B. P.; Merlitz, H.; Mahapatra, U. S.; Mukherjee, D. J. Chem. Phys. (in press).
- 14. Nayak, M. K. and Chaudhuri, R. K. (to be submitted).
- 15. Maroulis, G. and Bishop, D. M. J. Phys. B 1985, 18, 3653.
- 16. Maroulis, G. and Bishop, D. M. Chem. Phys. Letts. 1985, 114, 182.
- 17. Schmidt, M. W. and Ruedenberg, K. J. Chem. Phys. 1979, 71, 3951; Exponents for the Ar polarization functions are constructed from their p exponents.
- 18. Godbout, N.; Salahub, D.R.; Andzelm, J.; Wimmer, E. Can. J. Chem. 1992, 70, 560. Rb basis is generated by adding exponents to Godbout et al. basis set. The same basis set is used for Cs atom.