

# Gibbs Paradox and Stability of Porous Structure

Shu-Kun Lin

Molecular Diversity Preservation International (MDPI), Matthaeusstrasse 11, CH-4057 Basel, Switzerland; Tel. (+41) 79 322 3379; Fax: (+41) 61 302 8918; E-mail: lin@mdpi.org

When *Maurice A. Biot* developed the theory of fluid mixtures and stability of porous media, he recognized the difficulties posed by the Gibbs Paradox. Three of his papers are closely related to the resolution of Gibbs Paradox [1-3]. He is also well-known for the work on variational principles in thermodynamics [4]. When a porous medium is subjected to thermodynamic studies, "containers" of the fluids must be considered as part of the system under study, but thermodynamics treat the fluid (gases, liquids or solutions) body as the concerned system. The heat engine system considered in thermodynamics does not include the container. What is worse, Gibbs Paradox [5] states that the separation of the porous medium to form a bulky fluid phase and a pure bulky solid phase would not lead to any change in thermodynamic parameters; these two structures are of the same stability, which, intuitively, must be wrong. Recently carried out active studies on Gibbs Paradox [6-8] shed some light on this fundamental problem and, if agreed and accepted, the conclusions can be used as a theoretical foundation to study the stability of porous structures.

**Keywords:** information theory, entropy, deformation, mixing, separation (demixing)

## References and Notes

1. Biot, M.A. New Fundamental Concepts and results in thermodynamics with chemical applications. *Chemical Physics* **1977**, *22*, 183-198. (Paper 156 at <http://www.pmi.ou.edu/Biot2005/biotConferenceBiotsPapers.htm>).
2. Biot, M.A. Variational-Lagrangian irreversible thermodynamics of initially-stressed solids with thermomolecular diffusion and chemical reactions. *Journal of the Mechanics and Physics of Solids* **1977**, *25*, 289-307, and Errata, vol. 26, pp. 59, 1978. (Paper 157 at <http://www.pmi.ou.edu/Biot2005/biotConferenceBiotsPapers.htm>).
3. Biot, M.A. New Variational-Lagrangian Thermodynamics of Viscous Fluid Mixtures with Thermomolecular Diffusion. *Proceedings of the Royal Society of London. Series A, Mathematical and Physical Sciences* 1979, *365*, 467-494. (Paper 163 at <http://www.pmi.ou.edu/Biot2005/biotConferenceBiotsPapers.htm>).
4. See a list of papers at <http://www.mdpi.org/lin/09meetings/biot/biot-papers.htm>.
5. Gibbs paradox literature: <http://www.mdpi.org/lin/entropy/gibbs-paradox.htm>.
6. Lin, S.-K. Correlation of entropy with similarity and symmetry. *J. Chem. Inf. Comp. Sci.* **1996**, *36*, 367-376.
7. Lin, S.-K. Understanding structural stability and process spontaneity based on the rejection of the Gibbs paradox of entropy of mixing. *Theochem–J. Mol. Struc.* **1997**, *398*, 145-153.
8. Lin, S.-K. [Gibbs Paradox and the Concepts of Information, Symmetry, Similarity and Their Relationship](http://www.mdpi.org/lin/entropy/gibbs-paradox.htm). *Entropy* **2008**, *10*, 1-5. DOI: [10.3390/entropy-e10010001](https://doi.org/10.3390/entropy-e10010001). arXiv:[0803.2571](https://arxiv.org/abs/0803.2571).