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N,*N*'-Substituted 1,2,5 Thiadiazolidine 1,1-Dioxides: Synthesis, Selected Chemical and Spectral Proprieties and Antimicrobial Evaluation[†]

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[†]Dedicated to the memory of Professor Arrar Mostefa (1961-2004)

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Abstract: The sulfamide functional group is increasingly relevant in both medicinal and bioorganic chemistry. We report here practical access to a series of N2,N5-substituted five-membered cyclosulfamides. The five-membered heterocyclic motif was prepared starting from proteogenic amino acids and chlorosulfonyl isocyanate via the Mitsunobu reaction. Selected chemical and spectral proprieties and the antimicrobial evaluation of these compounds are detailed.

Keywords: Amino acids, cyclic sulfamides, cyclization, Mitsunobu reaction, propionylation, constrained peptides.

Introduction

The synthesis and reactivity of heterocyclic compounds containing sulfonyl moieties have attracted much interest in recent years because of the interesting chemical and biological proprieties associated with their structural similarities with biomolecules containing carbonyl groups. Cyclosulfamides have enjoyed popularity in the field of medicinal chemistry as nonhydrolysable components in peptidomimetics [1-2], agonists of the 5-HT_{ID} receptor (regulating serotonin levels) [3], HIV and serine protease inhibitors [4-7], and constrained di-and tripeptides [8]. The reported strategies for the synthesis of cyclosulfamides are based either on the incorporation of the sulfamoyl moiety by reacting

useful starting points for the construction of an array of peptidomimetic scaffolds and constrained di and-tripeptides. CSI has been found to be versatile reagent of great interest in synthetic heterocyclic chemistry [16]. In this case, CSI contains the required sulfonyl group and one of the nitrogens of the 1,2,5-thiadiazolidine1,1-dioxides.

Figure 1. Cyclosulfamide stuctures



In continuation of our efforts to design and synthesize new cyclic sulfamides, we have extended our studies to a series of new heterocyclic constrained peptides containing sulfamide groups C and D (Figure 1). The derivatization of amino acids allowed the introduction *N*- C^* moieties with a well-defined configuration. Herein, we describe the synthesis and the preliminary results of the biological evaluation of a series of these new heterocycles containing sulfamido groups.

Results and Discussion

As outlined in Scheme 1, the different heterocyles **1b-5b** were prepared in a two-step reaction sequence starting from (*tert*-<u>butyloxycarbonylsulfonyl</u>) L-amino acid methyl esters **1-5**. These compounds were synthesized by sulfamoylation of aminoester derivatives (Ala, Val, Leu, Asp, Glu) as previously described [17-19].

Scheme 1. General synthesis of N^2 , N^5 cyclosulfamides



Reagents and conditions: (a) Chloroethanol (1 equiv.), PPh₃ (1 equiv.), DEAD (1 equiv.), THF; (b) K₂CO₃ (1.5 equiv.), DMSO.

In these Boc-sulfamides 1-5, the Boc (*t*-butyloxycarbonyl) group increases the acidity of the adjacent NH group and allows an expedient regiospecific alkylation under Mitsunobu conditions [20-21] using chloroethanol, which provides the *N*-substituted Boc–sulfamides **1a-5a** in good yields. The cyclization reaction of these N,N-sulfamides **1a-5a** under basic conditions in DMSO gives N^2,N^5 -substituted cyclosulfamides **1b-5b** in satisfactory yields.

Selective cleavage of the *t*-butyloxycarbonyl protective group with trifluoroacetic acid gives compounds **1c-5c** in good yield (Scheme 2). N^2 , N^5 -Cyclosulfamides **1d-5d** were readily prepared in quantitative yield from the cyclosulfamides **1c-5c** by treatment with propionyl chloride in the presence of triethylamine. These compounds can be used in asymmetric aldol reactions. Also, attempts to incorporate the amino acid moiety employing the Mitsunobu reaction using an α -hydroxy ester (L-(-)-ethyl lactate) allowed us to obtain two constrained dipeptidal cyclic sulfamides **1e-2e** in moderate yields with inversion of the configuration.





Reagents and conditions: (a) TFA, CH₂Cl₂; (b) propionyl chloride (1 equiv.), TEA (1 equiv.), CH₂Cl₂; (c) (L)-(-) ethyl lactate (1 equiv.), PPh₃ (1 equiv.), DEAD (1 equiv.), THF.

The structures of all compounds were unambiguously confirmed by the usual spectroscopic methods: ¹H- and ¹³C-NMR spectroscopy, mass spectrometry and IR spectra.

Biological Activity

The bacterial strains used in this study were *Staphylococcus aureus* and *Escherichia coli* species. They were isolated from an aquatic medium, followed by successive isolations carried out periodically in specific media in order to obtain strains as pure as possible. The solid media of MacConkey and Chapman have been used for *Escherichia coli* and *Staphylococcus aureus*, respectively. Microscopic study, after Gram coloration, was carried out after incubation at 37°C for 24 hours. The biochemical characteristics of each strain have been determined using a classic biochemical gallery. Finally, the pathogenic power of *Staphylococcus aureus* has been confirmed by showing that the coagulase of this strain was hemolytic, *in vitro*, towards rabbit or human plasma.

To draw the antibiogram, the dilution in a liquid medium method was chosen. It is based on putting innoculums of each studied strain in contact with increasing concentrations of the cyclosulfamides **4b** and **4d**. In a glucose medium, each bacterial inoculum (100 μ L per suspension) was distributed in a series of tubes (macro-dilution method) containing increasing sulfamide concentrations [22]. The bacterial innoculum corresponding to the two studied strains, was previously prepared from a colony that was collected from a solid medium and then put in suspension in a glucose medium for 18 hours at 37°C. After the incubation of the whole tubes at 37°C for 24 hours, the MIC (minimal inhibited concentration) of each of the two cyclosulfamides with respect to each strain (*Escherichia coli* and/or *Staphylococcus aureus*) was measured as indicated by the tube that contained the lower concentration of the sensitivity of the bacteria *Escherichia coli* and *Staphylococcus aureus* towards cyclosulfamides **4b** and **4d** are presented in Tables 1 and 2, respectively.

μg/mL												
strain	2	3	5	8	10	15	20	30	40	50	80	100
E. coli	+	+	+	MIC	-	-	-	-	-	-	-	-
S. aureus	+	+	+	+	+	+	+	+	+	+	+	+

Table 2. Strain sensitivity towards cyclosulfamide 4d

μg/mL												
strain	2	3	5	8	10	15	20	30	40	50	80	100
E. coli	+	MIC	_	-	_	_	-	-	-	-	-	_
S. aureus	+	+	+	+	+	+	+	+	+	+	+	+

No inhibitory effect on the growth of the *Staphylococcus aureus* strain was observed in the presence of cyclosulfamide **4b**. The cyclosulfamide **4d**, also tested in this study, similarly showed no significant antimicrobial activity towards *S. aureus*. It is therefore clear that this species is resistant to these particular compounds. In contrast, however, a significant bacteriostatic effect has been observed towards *Escherichia coli*, with no growth being observed in tubes containing sulfamide concentrations

of **4b** equal or greater than 8 μ g/mL, while cyclosulfamide **4d** also showed a marked antimicrobial activity with regards to *Escherichia coli* and a MIC of 3μ g/mL was obtained for this molecule.

Conclusions

We have established a new synthetic strategy to prepare peptidic structures constrained with a cyclosulfamide moiety. The N^2 , N^5 -unsymmetric cylic sulfamides can be prepared in three steps (alkylation, cyclization, deprotection). We have also demonstrated the useful application of these cyclic sulfamides in the preparation of pseudopeptides. The preliminary results of antimicrobial activity are encouraging. Further biological evaluation of the resulting compounds and their incorporation into biomolecule analogues are currently in progress.

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Experimental

General

All commercial chemicals and solvents were used as received. Melting points were determined in open tubes on a Büchi apparatus and are uncorrected. IR spectra were recorded on a Perkin-Elmer spectrophotometer. Microanalyses were performed in the Microanalysis Laboratory of ENSCM (Montpellier). ¹H and ¹³C- Nuclear Magnetic Resonance spectra were determined on a Brüker AC 250 spectrometer. Chemical shifts are recorded in ppm (δ) and coupling constants in Hertz, relative to tetrametylsilane used as internal standard. Multiplicity is indicated as s (singlet), d (doublet), q (quadruplet), m (multiplet) and combinations of these signals. Fast-atom bombardment mass spectra (FAB) were recorded in positive or negative mode with glycerol (G), thioglycerol (GT), or 3-nitrobenzyl alcohol (NOBA) as matrix. Optical rotations for solutions in CHCl₃ were measured with a POLAX model 2L digital polarimeter. All reactions were monitored by Thin Layer Chromatography (TLC) on silica gel Merck 60 F₂₅₄ precoated aluminium plates, developed by spraying with ninhydrin solution. Column chromatography was performed using silica gel 60 (230-400 mesh).

General synthetic procedure for carbamoylation–sulfamoylation:

A solution of N-chlorosulfonyl *tert*-butylcarbamate (0.05 mol) was prepared by addition of *tert*butanol (4.8 mL in 50 mL of dried dichloromethane) into a solution of CSI (7.1 g in the same solvent). The resulting solution of Boc-sulfamoyl chloride (25 mL) and triethylamine (17.40 g, 17.1 mL, 85 mmol) in dichloromethane (100 mL) was added into a suspension of amino ester (0.05 mol) in the same solvent (120 mL) at 0°C. The reaction was complete in 45 minutes. The reaction mixture was then diluted with dichloromethane (100 mL) and washed with two portions of 0.1 N HCl solution. The organic layer was dried (Na₂SO₄) and concentrated *in vacuo* to give the crude product, which was purified by column chromatography eluting with dichloromethane to give compounds **1-5** in 75-80% yield. The synthesis by this general method of compounds **1-4** from CSI, *tert*-butyl alcohol and the methyl esters of the amino acids L-alanine, L-valine, L-leucine and L-aspartic acid was previously described [18-19].

(*S*)-*Dimethyl* [*N*-(*N*- tertiobutyloxycarbonyl)-sulfamoyl] glutamate (**5**). Yield=76%; TLC: R_f =0.34 (CHCl₂-MeOH 9:1); m.p=105-106°C; $[\alpha]_D$ =+17 (c=1; MeOH). IR (KBr) y cm⁻¹: 1744, 1706 (C=O), 1380 and 1160 (SO₂), 3305 and 3353 (NH). ¹H-NMR (CDCl₃) δ ppm: 7.72 (s, 1H), 5.55 (d, J=8.4 Hz, 1H), 4.42 (m, 1H), 3.72-3.80 (2s, 6H), 3.00 (ddd, J=16.5, 6.4, 6.4 Hz, 2H), 3.20 (ddd, J=16.5, 6.4, 6.4 Hz, 2H), 2,20 (m, 2H), 1.50 (s, 9H); ¹³C-NMR (CDCl₃ δ ppm: 173, 171, 151, 85, 56, 53, 52, 48, 41, 29, 28, 27; M.S: (NOBA, FAB< 0): 353 [M-H]⁻, 253, 707. M=354; Anal. Calcd. for C₁₂H₂₂ N₂O₈S: C, 40.67; H, 6.21; N, 7.90; S, 9.03; found C,40.73; H, 6.24; N, 7.94; S, 8.92.

General procedure for the synthesis of N-Boc, N-(2-chlroalkyl) sulfamides 1a-5a

A solution of *N-tert*-butyloxycarbonylsulfamoylamino esters **1-5** (30 mmol), triphenyl-phosphine (7.6 g) and chloroethanol (2.4g; 2 mL) in THF (25 mL) was added dropwise (20 min, 5°C) to a solution of equimolar quantities of diethyl (diisopropyl) azodicarboxylate (30 mmol; 5.22 g or 6.06 g) in the same solvent (25 mL). The reaction medium was stirred under an atmosphere of dry nitrogen for about 45 min. TLC revealed that the substituted compound formed is less polar than its precursor (UV, ninhydrin). Oxydoreduction compounds were removed by filtration after precipitation into diethyl ether. The filtrate was concentrated and the crude residue was purified by column chromatography eluting with dichloromethane. Substituted compounds **1a-5a** were recovered in 65-75% yield.

(*S*) *Methyl* [*N*,*N*'-*tert-butyloxycarbonyl*, *N*'-*chloroethyl*)*sulfamoyl*] *alaninate* (**1a**). Yield=65%; TLC: R_f =0.52 (CHCl₃); m.p.=88-90 °C; [α]_D = -17 (c=1, MeOH); IR (KBr) y cm⁻¹: 1744, 1706 (C=O), 1380 and 1160 (SO₂), 3305 (NH); ¹H-NMR (CDCl₃) & ppm: 5.55 (d, J=8.42 Hz, 1H), 4.42 (m, 1H), 3,70 (s, 3H), 3,65 (t, J=6.8 Hz, 2H), 3.95 (t, J=6.8 Hz, 2H), 1.50 (s, 9H), 1.30 (d, J=7.2 Hz, 3H). ¹³C-NMR (CDCl₃ & ppm: 170, 150, 84, 54, 52 49, 29, 27, 25; M.S: (NOBA, FAB<0): 343 [M-H]⁻, 243, 687. M=344-346; Anal. Calcd. for C₁₁H₂₁N₂O₆SCl: C, 38.31; H, 6.09; N, 8.12; S, 9. 28; found: C, 38.33; H, 6.09; N, 8.09; S, 9.22.

(*S*) *Methyl* [*N*-(*N*'-*tert-butyloxycarbonyl*, *N*'-*chloroethyl*)*sulfamoyl*] *valinate* (2a). Yield=68%; TLC: R_f =0.65 (CHCl₃); m.p.=60-62 °C; $[\alpha]_D$ =-9 (c=1; MeOH) ; IR (KBr) y cm⁻¹: 1730, 1700 (C=O), 1390 and 1160 (SO₂), 3310 (NH); ¹H-NMR (CDCl₃) § ppm: 5.80 (d, J=8.42 Hz, 1H), 3.95 (t, J=6.8 Hz, 2H), 3.87 (m, 1H), 3.68 (s, 3H), 3.65 (t, J=6.8 Hz, 2H), 2.10 (m, 1H), 1.52 (s, 9H), 0.98 (2d, J=6.9 Hz, 6H, 2CH₃); ¹³C-NMR (CDCl₃) § ppm: 172, 152, 85,55, 52, 48, 41, 28, 27, 22; M.S: (NOBA, FAB>0): 373 [M+H] ⁺, 273, 747. M=372-274; Anal. Calcd. for C₁₃H₂₅ N₂O₆SCl: C, 41.88; H, 6.71; N, 7.51; S, 8.59; found: C, 41.93; H, 6.74; N, 7.46; S, 8.51.

(*S*) *Methyl* [*N*-(*N*'-*tert*-*butyloxycarbonyl*, *N*'-*chloroethyl*)-*sulfamoyl*] *leucinate* (**3a**). Yield=75%; TLC: $R_f = 0.60$ (CHCl₃); m.p.=67-69°C; $[\alpha]_D = -23$ (c=1; MeOH); IR (KBr) y cm⁻¹: 1740, 1712 (C=O), 1390 and 1160 (SO₂), 3267 (NH); ¹H-NMR (CDCl₃) δ ppm: 5.80 (d, J=8.8 Hz, 1H), 3,95 (t, J=6.8 Hz, 2H), 3.87 (q, J=6.8 Hz, 1H), 3.68 (s, 3H), 3,65 (t, J=6.8 Hz, 2H), 1,87 (m, 2H), 1.55 (s, 9H), 1.48 (m, 1H), 0.98-0.88 (2d, J=3.9 Hz, 6H); ¹³C-NMR (CDCl₃) δ ppm: 172, 152, 85, 56, 53, 48, 43, 41, 28, 25, 24, 22; M.S: (NOBA, FAB>0): 387 [M+H]⁺, 287, 773. M=386-388; Anal. Calcd. for C₁₄H₂₇N₂O₆SCl: C, 43.46; H, 6.98; N, 7.24; S, 8.28; found: C, 43.49; H, 6.94; N, 7.20; S, 8.21.

(*S*) *Methyl* [*N*-(*N*'-*tert-butyloxycarbonyl*, *N*'-*chloroethyl*)-*sulfamoyl*] *aspartate* (**4a**). Yield=70%; TLC: $R_f = 0.67$ (CHCl₃); oil; $[\alpha]_D = -33$ (c=1; MeOH) ; IR (KBr) y cm⁻¹: 1755, 1715 (C=O) 1370 and 1130 (SO₂), 3269 (NH); ¹H-NMR (CDCl₃) δ ppm: 5.85 (d, J=8.4 Hz, 1H), 4.30 (q, J=8.4 Hz, 1H), 3.90 (t, J=6.7 Hz, 2H), 3.70-3.80 (2s, 6H), 3.68 (t, J=6.7Hz, 2H), 3.50 (2dd, J= 4.1, 8.4 Hz, 2H), 1.50 (s, 9H); ¹³C-NMR (CDCl₃ δ ppm: 171, 170, 150, 85, 55, 51, 52, 49, 42, 28, 27; M.S: (NOBA, FAB>0): 407 [M+H] ⁺, 307, 805. M=406-404; Anal. Calcd. for C₁₃H₂₃ N₂O₈SCl: C,38.75; H, 5.71; N, 6.95; S, 7.95; found: C, 38. 73; H, 5.74; N, 6.94; S, 7.92.

(*S*) *Methyl* [*N*-(*N*'-*tert-butyloxycarbonyl*, *N*'-*chloroethyl*)-*sulfamoyl*] glutamate (**5a**). Yield=72%; TLC: $R_f = 0.65$ (CHCl₃); oil; $[\alpha]_D = -21$ (c=1; MeOH) ; IR (KBr) y cm⁻¹: 1746, 1715 (C=O) 1395 and 1156 (SO₂), 3310 (NH); ¹H-NMR (CDCl₃) & ppm: 6.10 (d, J=8.5 Hz, 1H), 4.30 (m, 1H), 3.90 (t, J=6.7 Hz, 2H), 3.68 (t, J=6.9 Hz 2H), 3.60-3.65 (2s, 6H), 2.48 (m, 2H), 2.10 (m, 2H), 1.50 (s, 9H); ¹³C-NMR (CDCl₃ & ppm: 173, 171, 151, 85, 56, 53, 52, 48, 41, 29, 28, 27; M.S: (NOBA, FAB>0): 417 [M+H]⁺, 317, 833. M=416-418; Anal. Calcd. for C₁₄H₂₅ N₂O₈SCl: C, 40.33; H, 6.02; N, 6.72; S, 7.68; found: C, 40.42; H, 6.04; N, 6.75; S, 7.62.

General procedure for preparation of N-Boc-N-substituted)-1,2,5-thiadiazolidine 1,1-dioxides 1b-5b.

Cyclization with K₂CO₃ in DMSO

The 2-chloroalkyl compounds (10 mmol) were dissolved in dimethysulfoxide (DMSO) and anhydrous K_2CO_3 (1.5 equiv.) was added in one portion. The resulting mixture was stirred at room temperature for 8 h, diluted with dichloromethane (200 mL) and acidified with 5% HCl. The organic layer was washed with water, dried (Na₂SO₄) and concentrated under reduced pressure. The residue was purified by chromatography on silica gel. Recristallization of the crude product from CH₂Cl₂-petroleum ether (1:5) afforded the pure expected cyclosulfamides **1b-5b** in 82-92% yields.

 $(N^2-(2^{\circ}S)-Propionic acid methyl ester, N^5-tert-butyloxycarbonyl)-1,2,5-thiadiazolidine 1,1-dioxide ($ **1b** $). Yield=86%; TLC: R_f =0.58 (CHCl₃); Mp=132-134°C; [<math>\alpha$]_D=-34. (c=1; MeOH); IR (KBr) v cm⁻¹: 1750 1712, 1378 and 1167 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.15 (q, J=7.8 Hz, 1H), 3.80 (t, J=6.4 Hz 2H), 3.70 (s, 3H), 3.65 (t, J=6.4 Hz, 2H), 1.50 (d, J=7.8 Hz, 1H), 1.52 (s, 9H); ¹³C-NMR (CDCl₃ δ ppm: 172, 150, 84, 56, 52, 43, 39, 28, 27; M.S: (NOBA, FAB< 0): 307 [M-H]⁻, 207; M=308; Anal. Calcd. for C₁₁H₂₀N₂O₆S: C, 42.85; H, 6.49; N, 9.09; S, 10.39; found: C, 42.83; H, 6.44; N, 9.04; S, 10,32.

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 $(N^2-(2'S)-3'-methylbutyric acid methyl ester, N^5-tert-butyloxycarbonyl)-1,2,5-thiadiazolidine-1,1$ dioxide (**2b** $). Yield=90%; TLC: R_f =0.60 (CHCl₃); m.p.=154-155°C; [<math>\alpha$]_D=-38 (c=1; MeOH); IR (KBr) v. cm⁻¹: 1745, 1718 (C=O), 1390 and 1160 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.15 (d, J=7.2 Hz, 1H), 3,95 (t, J=6.4 Hz, 2H), 3.80 (t, J=6.4 Hz, 2H), 3.70 (s, 3H), 2.20 (m, 1H), 1.57 (s, 9H),1.30 (d, J=7.2 Hz, 3H), 0,98 (2d, J=6.9 Hz, 6H); ¹³C-NMR (CDCl₃ δ ppm: 172, 149, 84, 56, 53, 43, 39, 28, 26, 23, 22; M.S: (NOBA, FAB>0): 337 [M+H]⁺, 237; M=336; Anal. Calcd. for C₁₃H₂₄ N₂O₆S: C, 46.43; H, 7.14; N, 8.33; S, 9.52; found: C, 46.48; H, 7.17; N, 8.34; S, 9.44.

 $(N^2-(2'S)-4'-Methylpentanoic acid methyl ester, N^5-tert-butyloxycarbonyl)-1,2,5-thiadiazolidine-1,1$ dioxide (3b). Yield= 87%; TLC: R_f =0.58 (CHCl₃); m.p.=138-139 °C; [α]_D =-53 (c=1; MeOH) ; IR(KBr) y cm⁻¹: 1747, 1728 (C=O), 1360 and 1120 (SO₂) ; ¹H-NMR (CDCl₃) δ ppm: 4.30 (t, J=8.4 Hz),3.95 (t, J=6.4 Hz, 2H), 3.72 (s, 3H), 3.55 (t, J=6.4 Hz, 2H), 1.55-1.65 (m, 3H), 1.51 (s, 9H), 0.95-1.00(2d, J=6.9 Hz, 6H) ;¹³C-NMR (CDCl₃ δ ppm: 171, 149, 86, 54, 53, 43, 39, 37, 28, 25, 23, 21 ; M.S:(NOBA, FAB>0): 351 [M+H] ⁺, 251 ; M=350; Anal. Calcd. for C₁₄H₂₆ N₂O₆S: C, 48.00; H, 7.43; N,8.00; S, 9. 14; found: C, 48.07; H, 7.48; N, 7.94; S, 9.07.

 $(N^2-(2'S)-Bis(1',3'-methoxycarbonyl)ethyl),N^5$ -tert-butyloxycarbonyl)-1,2,5-thiadiazolidine-1,1-dioxide (4b). Yield=82%; R_f =0.61(CHCl₃); oil; [α]_D =+38 (c=1; MeOH); IR (KBr) <u>v</u> cm⁻¹: 1754, 1751, 1710 (C=O) 1390 and 1150 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.30 (m, 1H), 3.70-3.80 (2s, 6H), 3.95 (t, J=6.4 Hz, 2H, 3.55 (t, J=6.4 Hz, 2H), 3.50 (ddd, J=17.2, 7.1, 4.5 Hz, 2H), 1,50 (s, 9H); ¹³C-NMR (CDCl₃ δ ppm: 177, 171, 150, 85, 57 53, 52, 43, 39, 28, 25; M.S: (NOBA, FAB>0): 367 [M+H] ⁺, 267; M=366; Anal. Calcd. for C₁₃H₂₂ N₂O₈S: C, 42.62; H, 6.01; N, 7.65; S, 8.74; found : C, 42.63; H, 6.04; N, 7.69; S, 8.69.

(N^2 -(2'S')-Bis(1',4'-methoxycarbonyl)propyl, N^5 -tert-butyloxycarbonyl)-1,2,5-thiadiazolidine-1,1dioxide (**5b**). Yield=92%; TLC: R_f =0.58 (CHCl₃); oil; [α]_D=+38 (c=1; MeOH); IR (KBr) ½ cm⁻¹: 1748, 1745, 1720 (C=O), 1360 and 1120 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.30 (m, 1H), 3.95 (t, 2H, J=6,4 Hz), 3.68 (t, J=6,4 Hz, 2H), 3.72-3.65 (2s, 6H), 2.48 (m, 2H), 2.10 (m, 2H),1.50 (s, 9H); ¹³C-NMR (CDCl₃ δ ppm: 173, 170, 150, 85, 56, 53, 52, 43, 41, 30, 28, 27; M.S: (NOBA, FAB>0): 381 [M+H] ⁺, 281; M=380; Anal. Calcd. for C₁₄H₂₄ N₂O₈S : C, 44.21; H, 6.31; N, 7.37; S, 8.42; found C, 44.30; H, 6.34; N, 7.34; S, 8.36.

Deprotection

A solution of trifluoroacetic acid (50% in dried dichloromethane; 3 equiv) was added dropwise into a strirred solution of N,N'-substituted cyclosufamides **1b-5b** (20 mmol) in dried dichloromethane (15 mL) at 0°C. The reaction medium was stirred during two hours, concentrated under reduced pressure and coevaporated with diethyl ether. The residue was purified by flash chromatography. Elution with CH₂Cl₂-MeOH (95:5) gave deprotected cyclic sulfamides **1c-5c** in 85%-90% yield.

 N^2 -(2'S)-(*Propionic acid methyl ester*) 1,2,5-thiadiazolidine 1,1-dioxide (**1c**). Yield=90%; TLC: R_f =0.45 (CHCl₃); m.p.=124-125 °C; [α]_D=-34. (c=1; MeOH) ; IR (KBr) <u>y</u> cm⁻¹: 1751 (C=O) 1375 and 1160 (SO₂), 3345 (NH); ¹H-NMR (CDCl₃) δ ppm: 6.24 (t, J=6.7 Hz, 1H), 4.15 (q, 1H, J=7.8 Hz, 1H),

3,80 (t, J=6.7Hz, 2H), 3.70 (s, 3H), 3.65 (m, 2H), 1.50 (d, J=7.8 Hz, 3H); 13 C-NMR (CDCl₃ δ_{1} ppm: 172, 56, 52, 41, 39, 28; M.S: (NOBA, FAB< 0): 207 [M-H]⁻, 415; M=208; Anal. Calcd. for C₆H₁₂ N₂O₄S: C, 34.61; H, 5.77; N, 13.46; S, 15.38; found C, 34.63; H, 5.84; N, 13.44; S, 15.32.

 N^2 -(2'S)-(3-methylbutyric acid methyl ester) 1,2,5-thiadiazolidine 1,1-dioxide (**2c**). Yield=85%; TLC: R_f =0.55 (CHCl₃); m.p.=134-135°C; [α]_D =-44 (c=1; MeOH); IR (KBr) y cm⁻¹: 1745 (C=O), 1390 and 1160 (SO₂), 3340 (NH); ¹H-NMR (CDCl₃) δ ppm: 6.30 (tJ=6.8Hz, 1H), 4.15 (d, J=7.2 Hz, 1H), 3.90 (t, J= 6,7 Hz, 2H), 3.70 (s, 3H), 3.60 (m, 2H), 1.30 (m, 1H), 0,98 (2d, J=6.9 Hz, 6H); ¹³C-NMR (CDCl₃ δ ppm: 171, 56, 53, 41, 39, 26, 23, 22; M.S: (NOBA, FAB>0): 237 [M+H]⁺, 473; M=236; Anal. Calcd. for C₈H₁₆ N₂O₄S: C, 40.67; H, 6.78; N, 11.86; S, 13.60; found: C, 40.73; H, 6.84; N, 11.94; S, 13.52

*N*²-(2'*S*)-4'-methylpentanoic acid methyl ester) 1,2,5-thiadiazolidine 1,1-dioxide (**3c**). Yield= 90%; TLC: R_f =0.53 (CHCl₃); m.p.=128-130 °C; [α]_D=+76 (c=1; MeOH); IR (KBr) v cm-¹: 1747, 1360 and 1120 (SO₂), 3320 (NH); ¹H-NMR (CDCl₃) δ ppm: 6.30 (tJ=6.8Hz,1H), 4.30 (m, 1H), 3.85 (t, J=6.4 Hz, 2H), 3.72 (s, 3H), 3.55 (m, 2H), 1.55-1.65 (m, 3H), 0,98 (2d, J=6.9 Hz, 6H); ¹³C-NMR (CDCl₃ δ ppm: 177, 57, 53, 41, 39, 28, 25, 22, 21; M.S: (NOBA, FAB>0): 251 [M+H]⁺, 501. M=250; Anal. Calcd. for C₉H₁₈N₂O₄S: C, 43.20; H,7.20; N, 11.20; S, 12.80; found: C, 43.23; H, 7.24; N, 11.17; S, 12.72.

 N^2 -(2'S')-bis(1',3'-methoxycarbonyl)ethyl] 1,2,5-thiadiazolidine 1,1-dioxide (4c). Yield=88%; R_f =0.56 (CHCl₃); oil; [α]_D =-56 (c=1; MeOH); IR (KBr) y cm-¹: 1750-1755 (2C=O) 1390 and 1150 (SO₂), 3325 (NH); ¹H-NMR (CDCl₃) δ ppm: 6.80 (tJ=6.7Hz,1H), 4.25 (2d, J=7.1, 4.6 Hz, 1H), 3.70-3.80 (2s, 6H), 3.70 (t, J=6.7 Hz, 2H), 3.55 (m, 2H), 3.50 (ddd, J= J=17.2, 7.1, 4.5 Hz, 2H); ¹³C-NMR (CDCl₃ δ ppm: 177, 175, 57, 53, 52, 41, 39, 28; M.S: (NOBA, FAB>0): M=267 [M+H]⁺, 533; M=266; Anal. Calcd. for C₈H₁₄O₆N₂S: C, 36.09; H, 5.26; N, 10.52; S, 12.03; found: C, 36.13; H 5.24; N, 10.54; S, 12.00.

 N^{2} -(2'S')-bis(1',4'-methoxycarbonyl)propyl] 1,2,5-thiadiazolidine 1,1-dioxide (**5c**). Yield=89%; TLC: R_f =0.54 (CHCl₃); oil; [α]_{D=}+67 (c=1; MeOH) ; IR (KBr) y cm-¹: 1745-1738 (2C=O) 1360 and 1120 (SO₂), 3145 (NH) ; ¹H-NMR (CDCl₃) δ ppm: 6.60 (tJ=6.8Hz,1H), 4,30 (m, 1H), 3,70 (t, J=6.4 Hz, 2H), 3.65 (m, 2H), 3.72-3.65 (2s, 6H), 2.50 (m, 2H), 2.10 (m, 2H) ; ¹³C-NMR (CDCl₃ δ ppm: 172, 170, 56, 53, 52, 42, 40, 30, 28 ; M.S: (NOBA, FAB>0): 281[M+H] ⁺, 561. M=280; Anal. Calcd. for C₉H₁₆N₂O₆S: C, 38.57; H, 5.71; N, 10.00; S, 11.43; found: C,38.63; H, 5.74; N, 10.04; S, 11.36.

Propionylation

To a stirring solution of N-substituted cyclosulfamide 1c-5c (20 mmol), in dichloromethane (50 mL) was added triethylamine (1.1 equiv., 22 mmol, 2.22 g, 1.60 mL), and catalytic quantities of dimetylaminopyridine (DMAP). Propionyl chloride (1.5 equiv., 30 mmol, 2.41 g, 2.57 mL) diluted in the same solvent (15 mL) was added slowly to the resulting solution. When the addition was completed, the reaction mixture was stirred under an atmosphere of dry nitrogen. TLC reveals the formation of a substituted compound less polar than its precursor. The reaction mixture was concentrated *in vacuo*. The residue diluted with dichoromethane (50 mL), acidified with 0.1 N HCl solution and washed with water. The organic layer was dried with (Na₂SO₄) and concentrated under

reduced pressure to give the crude product. The residue was purified on silica gel by column chromatography eluting with dichloromethane to give the N^2 , N^5 substituted cyclosulfamides **1d-5d** in 75-90 % yields.

[N^2 -(2'S')-(propionic acid methyl ester), N^5 -propionyl] 1,2,5-thiadiazolidine 1,1-dioxide (1d). Yield=-90%; TLC: R_f=0.59 (CHCl₃); m.p.=88-89°C; [α]_D=-17 (c=1; MeOH); IR (KBr) y cm⁻¹: 1750-1715 (C=O), 1375 and 1160 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.15 (q, J=7.8 Hz, 1H), 3.80 (t, J=6.2 Hz, 2H), 3.70 (s, 3H), 3.65 (t, J=6.7 Hz, 2H), 2.85 (q, J=7.4 Hz, 2H), 1.50 (d, 3H, J=7.8 Hz), 1.15 (t, 3H, J=7.4 Hz, 3H); ¹³C-NMR (CDCl₃) δ ppm: 172, 170, 56, 53, 41, 39, 29, 28, 12; M.S: (NOBA, FAB>0): 265 [M+H] ⁺, 208; M=264; Anal. Calcd. for C₉H₁₆O₅N₂S: C, 40.91; H, 6.06;N, 10.60; S, 12.12; found: C, 40.98; H, 6.17; N, 10.65; S, 12.05.

[N^2 -(2'S)-(3'-methylbutyric acid methyl ester), N^5 -propionyl] 1,2,5-thiadiazolidine 1,1-dioxide (2d). Yield=88%; TLC: R_f =0,62 (CHCl₃); m.p.=94-95°C; [α]_D =-14 (c=1; MeOH). IR (KBr) y cm⁻¹: 1748-1712 (C=O), 1389-1163 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.15 (d, J=7.2 Hz, 1H), 3.95 (t, J=6.7 Hz, 2H), 3.8 (t, J=6,7 Hz, 2H), 3.70 (s, 3H), 2.85 (q, 2H, J=7.4 Hz, 2H), 1.30 (m, 1H), 1.16 (t, J=7,4 Hz, 3H), 0.98 (2d, J=6.9 Hz, 6H); ¹³C-NMR (CDCl₃ δ ppm: 172, 170, 56, 53, 43, 39, 26, 28, 23, 22, 13; M.S: (NOBA, FAB>0): M=373 [M+H]⁺, 174, 745; M=372; Anal. Calcd. for C₁₁H₂₀ N₂O₅S; C, 45.20; H, 6.85; N, 9.59; S, 10.96; found: C, 45.23; H, 6.19; N, 9.54; S, 10.89.

[N^2 -(2'S)-4'-methylpentanoic acid methyl ester), N^5 -propionyl] 1,2,5-thiadiazolidine 1,1-dioxide (**3d**). Yield=85%; TLC: R_f =0.65 (CHCl₃); m.p.=106-108°C; [α]_D=+54 (c=1; MeOH); IR (KBr) y cm⁻¹: 1747-1718 (C=O), 1362 and 1125 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.10 (m, 1H), 3.92 (t, J= 6.7 Hz, 2H), 3.75 (t, J=6.7 Hz, 2H), 3.72 (s, 3H), 2.85 (q, J=7,4 Hz, 2H), 1,55-1,65 (m, 12H), 1.15 (t, J=7.4 Hz, 3H), 0.98-1.00 (2d, J=6.9 Hz, 6H); ¹³C-NMR (CDCl₃ δ ppm: 175 ,170, 57, 53, 41,39, 28, 29, 25, 23, 21, 12; M.S: (NOBA, FAB>0): 307 [M+H]⁺, 250; M=306; Anal. Calcd. for C₁₂H₂₂N₂O₅S: C, 47.06; H, 7.19; N, 9.15; S, 10.46; found: C, 47.12; H, 7.25; N, 9.08; S, 10.42.

[N^2 -(2'S')-bis(1',3'-methoxycarbonylethyl), N^5 -propionyl] 1,2,5-thiadiazolidine 1,1-dioxide (4d). Yield=80%, TLC: R_f =0.67 (CHCl₃); oil; [α]_D =-87 (c=1; MeOH); IR (KBr) y cm⁻¹: 1749, 1753 and 1715 (C=O) 1380 and 1150 (SO₂); ¹H-NMR (CDCl₃) δ ppm: 4.30 (2d, J=7.3, 4.6 Hz, 1H), 3.70-3.80 (2s, 6H), 3.85 (t, J=6.8 Hz, 2H), 3.60 (t, J=6.8 Hz, 2H), 3.50 (ddd, J=J=17.2, 7.3, 4.6 Hz, 2H), 2.85 (q, J=7.4 Hz, 2H), 1.18 (t, J=7.4 Hz, 3H) ; ¹³C-NMR (CDCl₃ δ ppm: 173, 172, 165, 57, 53, 52, 43, 40, 29, 25, 12; M.S: (NOBA, FAB>0): M=323 [M+H] ⁺, 266. M=322; Anal. Calcd. for C₁₁H₁₈N₂O₇S: C, 40.99; H, 5.54; N, 8.69; S, 9.34; found: C, 41.03; H, 5.64; N, 8.80; S, 9.30.

[N^2 -(2'S')-bis(1',4'-methoxycarbonylpropyl), N^5 -propionyl] 1,2,5-thiadiazolidine 1,1-dioxide (5d). Yield=75%, TLC: R_f =0.51 (CHCl₃); oil; [α]_D=+43 (c=1; MeOH); IR (KBr) y cm-¹: 1745, 1738 and 1713 (C=O), 1360 and 1120 (SO₂); ¹H-NMR (CDCl₃) & ppm: 4.30 (m, 1H), 3.85 (t, J=6.4Hz, 2H), 3.65 (t, J=6.4 Hz, 2H), 3.72-3.65 (2s, 6H), 2.48 (m, 2H), 2.10 (m, 2H), 2.85 (q, J=7,4 Hz, 2H), 1.14 (t, J=7.4 Hz, 3H); ¹³C-NMR (CDCl₃) & ppm: 173, 171, 165, 56, 53, 52, 42, 41, 30, 29, 25, 12; Anal. Calcd. for C₁₂H₂₀O₇N₂S: C, 42.86; H, 5.95; N, 8.33; S, 9.52; found: C, 42.92; H, 9.87; N, 8.28; S, 9.43.

Alkylation via the Mitsunobu Reaction

To a stirring solution of N-substituted cyclosulfamide **1d-2d** (3.23 mmol) in THF (2 mL) was slowly added DEAD (3.23 mmol, 0.5 mL) via dropwise addition. A solution consisting of (L)-(-)- ethyl lactate (3.23 mmol, 0.37 mL) and PPh₃ (3.23 mmol, 847 mg) in THF (3mL), was slowly transferred via cannula into the cyclosulfamide solution. The reaction medium was stirred under an atmosphere of dry nitrogen for about 45 min. TLC reveals (UV, ninhydrin) the formation of a substituted compound less polar than its precursor. Oxydoreduction compounds were removed by filtration after precipitation into diethylether. The filtrate was concentrated and the crude residue was purified by columm chromatography eluting with dichloromethane. *N*,*N*^{*}-Substituted cyclosufamides **1e-2e** were recovered in 65-75% yield.

[N^2 -(2S)-(Methoxycarbonylethyl), N^5 -(2'R)-(propionic acid ethyl ester)]-1,2,5-thiadiazolidine 1,1dioxide (1e). Yield=65%; TLC: R_f=0.64 (CHCl₃); m.p.=88-89°C; [α]_D=-65 (c=1; MeOH); IR (KBr) y cm⁻¹: 1750-1731 (C=O), 1360 and 1152 (SO₂); ¹H-NMR (CDCl₃) & ppm: 4.76 (q, J=7.3, 1H), 4.65 (q, J=7.2, 1H), 4.15 (q, J=7.2 Hz, 1H), 3.80 (t, J=6.7 Hz, 2H), 3.70 (s, 3H), 3.65 (t, J=6.7 Hz, 2H), 1.44 (d, J=7.2, 3H), 1.39 (d, J=7.1 Hz, 3H), 1.28 (t, J=7.2, 3H); ¹³C-NMR (CDCl₃ & ppm: 172, 170, 56, 55, 53, 50, 41, 39, 29, 28, 12; M.S: (NOBA, FAB>0): M=309 [M+H] ⁺. M=308; Anal. Calcd. for C₁₁H₂₀ N₂O₆N₂S C, 42.86, H, 6.49, N, 9.09. S,10.03. Found: C, 40.86, H, 6.13, N,9.47. S, 10.03.

 $[N^2-(2S)-(3-Methylmethoxycarbonylpropyl), N^5-(2'R)-propionic acid ethyl ester)]$ 1,2,5-thiadiazolidine-1,1-dioxide (2e). Yield =75%; TLC: R_f =0.61 (CHCl₃); m.p.=80-81°C; [α]_D =-12.0 (c=0,5, CHCl₃); IR (KBr) y cm-¹: 1745, 1729, 1346, 1145; ¹H-NMR (CDCl₃) δ ppm: 4.70 (q, J=7.3 Hz, 1H), 4.20 (q, J=7.1Hz, 2H), 3.90 (d, J=3.3 Hz, 1H), 3.80-3.60 (m, 4H), 3.70 (s, 3H), 2.12 (m, 1H), 1.39 (d, J=7.1 Hz, 3H), 1.26 (t, J=7.3 Hz, 3H), 0.98 (d, J=6.8 Hz, 3H), 0.88 (d, J=6.9 Hz,3H); ¹³C-NMR (CDCl₃ δ ppm: 172, 169, 56, 55, 53, 52, 50, 41, 39, 28, 22, 21, 14; M.S: (NOBA, FAB>0): 337 [M+H]+. M=336; Anal. Calcd. for C₁₃H₂₄N₂O₆S: C, 46.43; H, 7.14; N, 8.33; C, 9.52. Found: C, 46.49; H, 7.18; N, 8.34; S, 9.43.

References

- 1. Groutas, W. C.; Kuang, R.; Venkataraman, R.; Epp, J.B.; Ruan, S.; Prakash, O. *Biochemistry* **1997**, *36*, 4739-4750.
- Groutas, W. C.; Kuang, R.; Ruan, S.; Epp, J. B.; Venkataraman, R.; Truong, T. M. *Bioorg. Med. Chem.* 1998, 8, 661-671.
- Castro, J. L.; Baker, L.; Guiblin, A. R.; Hobbs, S. C.; Jenkins, M. R.; Russel, M. G. N.; Beer, M. S.; Stanton, J. A.; Scholey, K.; Hargreaves, R. J. J. Med. Chem. 1994, 37, 3023-3032.
- Bäkbro, K.; Löwgren, S.; Österlund, K.; J. Atepo, J.; Unge, T.; Hultén, J.; Bonham, N. M.; Schaal, W.; Karlén, A. Hallberg, A. J. *J. Med. Chem.* 1997, 40, 898-902.
- 5. Hultén, J.; Bonham, N. M.; Nillroth, U.; Hansson, T.; Zucccarello, G.; Bouzide, A.; Aqvist, J.; Classon, B.; Danielson, U. H.; Karlèn, A. *J. Med. Chem.* **1997**, *40*, 885-897.
- Schaal, W.; Kalsson, A.; Ahlsén, G.; Andersson, H. O.; Danielson, U. H.; Classon, B.; Unge, T.; Samuelsson, B.; Hultén, A.; Hallberg, A. J.; Karlén, A. J. Med. Chem. 2001, 44, 155-164.

- (a) Lai, Z.; Gan, X.; Wei, L.; Alliston, K. R.;Yu,H.; Li, Y.H.; W. C. Groutas, W. C. Arch. Biochem. Biophys. 2004, 429, 191-197; (b) Groutas, W. C.; Epp, J. B.; Kuang, R.; Ruan, S.; Chong, S.L.; Venkataraman, R.; Tu, J.; He, S.; Fu,Q.; Y.H. Li.; Truong, T. M.; Vu. N. Arch. Biochem. Biophys. 2001, 385, 162-169.
- 8. Boudjabi, S.; Dewynter, G.; Voyer, N.; Toupet, L.; Montero, J.L *Eur. J. Org. Chem.* **1999**, 2275-2283.
- 9. M. Knollmuller, M. Monatsh. Chem. 1970, 101, 1443-1448.
- 10. Ahn, K. H.; Yoo, D. J.; Kim, J. S.; Tetrahedron Lett. 1992, 33, 6661-6664.
- Dougherty, J. M.; Probst, D. A.; Robinson, R. E.; Moore, D. J.; Klien, T. A.; Snelgrove, K. A.; Hanson, P. R.*Tetrahedron* 2000, *56*, 9782- 9790; (b) Jun, J. H.; Dougherty, J. M.; Probst, D. A.; Jiménez, M. S.; Hanson, P. R. *Tetrahedron* 2003, *59*, 8901-8912.
- 12. Johnson, P. D.; Jewell, S. A.; Romero, D.L. Tetrahedron Lett. 2003, 44, 5483-5485.
- 13. Regainia, Z.; Abdaoui, M.; Aouf, N.; Dewynter, G.; Montero, J. L. *Tetrahedron* **2000**, *56*, 381-387.
- 14. Regainia, Z.; Winum, J.Y.; Smain, F.T.; Toupet, L.; Aouf, N.; Montero, J. L. *Tetrahedron* **2003**, *59*, 6051-6056.
- 15. Berredjem, M.; Djebbar, H.; Regainia, Z.; Aouf, N.; G. Dewynter, G.; J.-Y. Winum, J. Y.; Montero, J. L. *Phosphorus, Sulfur Silicon* **2003**, *178*, 693-705.
- 16. Dhar, N.D.; Murthy, K. S. Synthesis 1986, 437-449
- 17. Dewynter, G.; Aouf, N.; Regainia, Z.; Montero, J.L. Tetrahedron 1996, 52, 993-1004.
- 18. Dewynter, G.; Aouf, N.; Criton, M.; Montero, J. L. Tetrahedron 1993, 49, 65-76.
- 19. Aouf, N.; Dewynter, G.; Montero, J. L. Tetrahedron Lett. 1991, 32, 6545-6546.
- 20. Mitsunobu. O. Synthesis 1981, 1-29.
- 21. Hughes, D. L. Org. React. 1992, 42, 335-380.
- 22. Laverdière, M.; Sabath, L.D. J. Med. 1977, 44, 73-88.

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