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# Toward a Library Synthesis of the Natural Dipeptide Antibiotic TAN 1057 A,B

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**Abstract**: The natural dipeptide antibiotic TAN 1057 A,B represents a promising new antibiotic entity. In this communication we report a novel approach for the synthesis of TAN 1057 A,B analogs bearing variations in the  $\beta$ -arginine side chain. This approach involves a combination of liquid and solid phase methods and allows for a library synthesis of analogs of the natural product.

Keywords: TAN 1057 A, B, dipeptide antibiotics, combinatorial chemistry.

# Introduction

The natural dipeptide antibiotic TAN 1057 A,B (**1**, Figure 1) was first isolated and characterized by Takeda [1]. To date, two independent total syntheses of this natural product have been disclosed [2]. TAN 1057 A,B displays excellent *in vitro* and *in vivo* activity against staphylococci including *Methicillin Resistant Staphylococcus Aureus* (MRSA) and represents a promising new antibiotic entity [3]. However, TAN 1057 A,B suffers from strong toxic side effects ( $LD_{50} = 50 \text{ mg/Kg}$ ) which are in our view prohibitive for therapeutic use in humans. Therefore, we launched a synthesis program with the goal of identifying

analogs with significantly improved tolerance and concomitant retention of the promising antibiotic activity. In this communication we report a new strategy that allows for the parallel synthesis of novel derivatives of the natural product focusing on alteration of the  $\beta$ -arginine moiety.

#### Figure 1: TAN 1057 A, B (1).



#### **Results and Discussion**

The salient molecular architecture of TAN 1057 A,B defines its unique physical and chemical properties: the natural product is highly water soluble, displays almost no retention on RP-HPLC material under standard conditions [4] and proves unstable towards nucleophiles and strong bases at elevated temperatures. This instability presumably results from the dihydropyrimidinone heterocycle which decomposes readily under the aforementioned conditions. However, early structure activity relationship (SAR) studies revealed that the heterocyclic core was essential for biological activity. Therefore, our program was primarily focused on alterations of the  $\beta$ -arginine side chain.

For the parallel synthesis of TAN 1057 analogs we envisioned a convergent strategy combining solid and liquid phase chemistry. Our synthesis commenced with the construction of amine **4** as outlined in Scheme 1. Accordingly, orthogonally protected  $\beta$ -lysine **2** was coupled with the dihydropyrimidinone core **3** [2b] employing standard EDC-peptide coupling conditions followed by HCl mediated deprotection of the terminal BOC-group.

Scheme 1: Synthesis of amine 4.



For the requisite installation of the guanidine moiety we employed the method published by *Dodd* and *Wallace* [5] who described the use of immobilized isothioureas **6** as convenient guanylation agents (cf. Scheme 2). Isothioureas **6** are readily available by a Mitsunobu reaction starting from commercially available resin **5**.

For method evaluation we selected six structurally diverse primary alcohols  $R_1OH$  (see Table 1) and submitted them to the Mitsunobu protocol as outlined in Scheme 2. Subsequently we used these polymeric intermediates **6** without further characterization to install the guanidine moiety at the terminus of **4**. To our dismay, the conditions originally described to effect guanylation (excess amine, 50°C, DMF) failed in our hands to deliver protected intermediates **7**. This was presumably due to the inherent instability of the heterocyclic moiety towards the given reaction conditions.

Scheme 2: Parallel synthesis of TAN 1057 A,B analogs.



Thus, we surveyed a range of additional bases such as ethyl diisopropylamine, N-methylmorpholine, DBU, Schwesinger type bases or proton sponges to promote the desired transformation under milder conditions. Finally, the optimal conditions for the guanylation included the use of a slight excess of resin **6** with respect to the hydrochloride **4**, DMF as solvent of choice, and DBU as base at room temperature for 3 days. The addition of transition metal salts like silver triflate or mercury chloride did not enhance the reaction rate significantly.

Subsequently, with an optimized coupling protocol in hand, all six polymeric isothioureas 6 were each reacted with 4 in a parallel setup. LC/MS analyses of the crude reaction mixtures revealed that protected intermediates 7 were formed as main products in all cases, however, the crude mixtures also contained byproducts arising from degradation of the dihydropyrimidinone. Purification of the crude products was achieved by automated preparative HPLC to deliver a range of modified natural product precursors 7 in acceptable yields around 40% (over 2 steps starting from 5) and sufficient HPLC-purities (>90% in all

cases). Finally, the target molecules **8** were obtained from **7** upon deprotection of the amino groups using HBr in acetic acid followed by precipitation of the products from acetone-methanol mixtures.

The results for all TAN 1057 A,B analogs are depicted in Table 1. In general, the overall yields (21 - 32%, 3 steps from 5) were acceptable taking into account that the dihydropyrimidinone core proved susceptible towards decomposition in the guanylation event. Moreover, HPLC analyses revealed that the majority of products were isolated in essentially pure form.

entry	alcohol R <sub>1</sub> OH	product <sup>a</sup>	overall yield from <b>5</b> (%) <sup>b</sup>	HPLC purity (%)⁵
1	ОН	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array} \begin{array}{c} \end{array}\\ \end{array} \begin{array}{c} \end{array}\\ \end{array} \begin{array}{c} \end{array} \begin{array}{c} \end{array}\\ \end{array} \begin{array}{c} \end{array}\\ \end{array} \begin{array}{c} \end{array} \begin{array}{c} \end{array}\\ \end{array} \begin{array}{c} \end{array} \end{array} \begin{array}{c} \end{array} \begin{array}{c} \end{array} \end{array} $	32	94
2	ОН	<b>8a</b> Me NH NH NH <sub>2</sub> O N N NH <sub>2</sub> HN NH NH <sub>2</sub> O N N NH O	30	97
3	∆он	$\begin{array}{c} \mathbf{BD} \\ \mathbf{N} \\ \mathbf$	26	88
4	OAc	$\begin{array}{c} OAc \\ H \\ $	22	80
5	OMe OH	8d Me OMe H NH NH NH <sub>2</sub> O N NH NH <sub>2</sub> NH NH <sub>2</sub> NH NH NH NH NH NH NH NH NH NH	21	84
6	NHBOC ОН	NH <sup>2</sup> H NH NH <sub>2</sub> O N H NH <sub>2</sub> HN Bf He	26	97

### Table 1: Analogs of TAN 1057 A,B by parallel synthesis.

(a) All products were obtained as HBr salts, (b) see Experimental section for details.

#### Conclusions

We have demonstrated an efficient and convergent route to TAN 1057 A,B analogs via a parallel synthesis approach. Key features of this strategy include the use of a diverse set of polymer bound isothioureas originating from a solid phase Mitsunobu protocol, optimized guanylation conditions, automated HPLC purification of the protected intermediates, and elaboration of a crystallization protocol for the final products. An application of this methodology for multi-component library synthesis and in depth pharmacological investigations of the synthesized compounds are currently underway.

#### **Experimental**

General procedure for the synthesis of 8 (90  $\mu$ mol scale) as outlined in Scheme 2: To a suspension of 1.5 eq of functionalized resin 5 (purchased from *Novabiochem*) in dry THF (3 mL), triphenylphosphine (9.4 eq) and the corresponding alcohol (9.4 eq) were added. Subsequently DIAD (9.0 eq.) was added dropwise and the resulting orange suspension was shaken at room temperature overnight. The solvent was filtered off and the resin was thoroughly washed with DMF (4 x 5mL), THF (3 x 5mL), MeOH (3 x 5mL) and dichloromethane (3x 5 mL).

The functionalized resin **6** thus obtained was suspended in DMF (2.5 mL) and the hydrochloride **4** (1.0 eq) was added in one portion. Finally, DBU (5 eq) was also added dropwise and the suspension was shaken at room temperature for 3 days. After filtration the resin was rinsed with DMF and the solvent was removed *in vacuo*. The crude products were purified using a Gilson automated preparative HPLC (conditions: Kromasil 100 C18, 5  $\mu$ m, 50 x 20 mm; gradient: acetonitrile / water 15:85 to 90:10, room temperature, flow 25 mL / min, detection @ 220 nm) to yield **7** (35-45% yield, >90% HPLC purity in all cases).

Finally, at 0°C HBr (30% solution in acetic acid, 1 mL) was added to a solution of **7** in dichloromethane (2 mL) and the mixture was stirred at 0°C for 1 hour. The solvent was removed *in vacuo*. In the final step the products were redissolved in methanol (0.5 mL) and acetone was added until the hydrobromide salts precipitated from the solution. After filtration compounds **8** were obtained as white powders. The products were analyzed by HPLC (conditions: Kromasil 100 C18, 25 x 4 mm; gradient: A = water containing PIC B7 low UV, B = acetonitrile, 0-1 min 90% A, 1-9 min 90% A to 90% B, 9-13 min 90% B; room temperature, flow 2 mL / min, detection @ 210 nm), <sup>1</sup>H-NMR (D<sup>4</sup>-methanol) and MS.

#### **References and Notes**

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Sample Availability: Samples of entries 1, 5, 7 and 11 are available from MDPI

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