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Synthesis and 1,3-Dipolar Cycloaddition Reactions of Chiral Maleimides

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Abstract: New routes to the synthesis of various novel chiral maleimides are described. The oxabicyclic anhydride **2** readily available *exo*-Diels-Alder adduct of furan and maleic anhydride was used as a vehicle, which in turn reacted with hydrochlorides of amino acids **3a-f** in the presence of Et_3N with release of furan to give the requisite novel chiral imides **4a-f** in good to moderate yields. The stereoselectivity of 1,3-dipolar cycloaddition of nitrile oxides with prepared chiral imides **4a-f** is investigated.

Keywords: Chiral maleimides, 1,3-dipolar cycloadditions, nitrile oxides, stereoselectivity.

Introduction

Maleimides are an important class of substrates for biological and chemical applications. In biological applications they are used as chemical probes of protein structure [1], as immunoconjugates for cancer therapy, as solid supported enzymes for synthetic applications, as haptene for the production of antibodies [2] or as new herbicides and pesticides [3]. Some compounds of the dicarboxyimide type are reported to reveal effective systeme activity against Botrytis cinerea, Cochliobolus miyabeanus and Pellicularia sasaci [3]. N-(3,5-Dichlorophenyl)pyrrolidine-2,5-dione (Dimetachlon) is being used as a protective and curative fungicide [4] and the cycloadduct of N-(3,5-dichlorophenyl)maleimide to furan [5] and its derivatives have also considerable fungicidal properties. The maleimide moiety can be used as a platform in synthesis due to its Michael accepting ability, dienophilic nature [6,7] as well as to its reactivity as a dipolarophile in 1,3-dipolar cycloadditions [8].

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Despite this, there are few reports in the literature of the synthesis of *N*-substituted chiral maleimides. Most methods involve the reaction of an amine with maleic anhydride, followed by dehydration of the intermediate maleamic acid, usually promoted by acids [9]. This method is limited to the use of amines as starting material and excludes those, that are unstable to the dehydration conditions.

In our effort to investigate the stereoselectivity of cycloaddition reactions to heterocyclic compounds, we have paid attention to the simple and effective synthesis of novel chiral dipolarophiles. Biagini and co-workers [10] recently published a simple synthesis of amino acids derivatives **1**, which contains in their skeleton the norbornenyl function. The authors describe two methods: a) a high temperature (one-pot) procedure and b) a low temperature (two steps) rout [10]. The second, milder method, allowed the synthesis of rather unstable products. All of the norbornens **1** prepared in this way were optically active, providing the first indication that the

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relatively harsh reaction conditions had not caused complete racemisation of the amino acids -centre. On the basis of this report we have started to test some possibilities to find efficient methods to prepare the *N*-substituted chiral maleimides (Scheme 1).



a: $R = CH_3$, b: $R = (CH_3)_2CH$, c: $R = (CH_3)_2CHCH_2$, d: $R = HOCH_2$, e: $R = CH_2COOCH_2CH_3$, For : **4a,b,d,e** $Z = CH_3CH_2$, **4c** $Z = CH_3$

4f

()



COOMe

Results and Discussion

We chose the oxabicyclic anhydride 2, a readily available *exo*-Diels-Alder adduct of furan and maleic anhydride [11] as a vehicle, which in turn reacted with

hydrochlorides of amino acids **3a-f** in the presence of Et_3N with release of furan to give requisite novel chiral imides **4a-f** in good to moderate yield (40-60%, Scheme 2). Yields, reported for the detailed application of the Biagini method [10] (1 eq. of amino acids, 0.9 - 1 eq. of

oxabicyclic anhydride **2**, 1.2 - 1.7 eq. of Et_3N , 20 - 29 h) were in all cases low (Table 1, Method A) and for (*S*)-alanine **4a** and (*S*)-valine **4b** derivatives this synthetic

route was not successful. Only signals for the uncyclized form were detected in the ¹H NMR spectra after work-up.



Scheme 2. Synthesis of maleimides 4a-f.

Therefore we had to change this procedure. The prolongation of the reaction time (19-56 h, TLC monitor- ing) and the use of excess Et₃N (1.7–2 eq. of amino acids) resulted in increased yields. (Table 1, Method B). In contrast to the previously mentioned Method A, after completion of the reaction, the reaction mixture was purified by column chromatography. Thus, the yields of all derivatives were increased and we were also able to isolate (*S*)-alanine **4a** and (*S*)-valine **4b** maleimides, in moderate yield. This modification was also successful for the heterocyclic amino acid - (*S*)-tryptophane **4f**.

We performed this modified synthetic route for the direct synthesis of required compounds by the reaction of maleic anhydride with the hydrochloride salts of amino acids. In the cases of (S)-alanine **4a** and (S)-valine **4b** the high temperature procedure did not give the expected *N*-maleonyl derivative, but rather the intermediate amino acids were obtained. On the other hand, in the case of (S)-leucine and (S)-aspartic acid salts the substituted maleimides were isolated in spectroscopically pure form (Method C, Scheme 3).



Scheme 3. Synthesis of maleimides 4a-f by method C.

Compd.	Method A	Method B	Method C
	Yield (%)	Yield (%)	Yield (%)
4 a	< 5	28	< 5
4b	< 5	50	< 5
4c	24	44	49
4d	16	Х	< 5
4e	39	62	55
4f	Х	50	Х

 Table 1. Yields of required maleimides related to synthetic method.

x - this method was not tried



Scheme 4. Formation of 4a-f and 5a-f.

As an additional product chiral fumaric acids *N*-substituted amides **5** were formed (60-20%) in all cases. The structure of (*E*)-**5** was confirmed by ¹H NMR spectroscopy. The value of the interaction constant *J* 14-16 Hz was determined and clearly proved the *trans* arrangement of the olefinic protons.

We suppose, that at high temperature two competitive reactions are possible (Scheme 4). Uncyclized maleamic 6 derivative can cyclize, the corresponding oxabicyclic derivative can undergo a *retro*-Diels-Alder reaction, furan is evaporated and the required product 4 is formed. The intermediate 6 can izomerize to *trans*-6 to give the more stable *E*-isomer 5.

In an effort to prevent these concurrent reactions, we performed the reaction of **2** with the (*S*)-aspartic derivative in various solvents. Analysis of the NMR spectra of the crude reaction mixture showed, that in the case of CH_2Cl_2 as a solvent only signals for the uncyclized derivative **6** as well as signals for the starting bicyclic derivative **2** were detected and maleimide **4** was not formed. After heating in benzene only signals for the uncyclized intermediate **6** were confirmed. The useful synthetic scope of this reaction for the preparation of maleimide **4** is given at temperature over 90°C.

Some 1,3-dipolar cycloaddition reactions were than tested with the prepared imides with the aim of investigating the stereocontrol of these additions. Two dipoles were chosen, the stable 2,4,6-trimethylphenylnitrile oxide [12] reacted by heating in benzene as well as benzenenitrile oxide generated from the corresponding benzhydroxymoyl chloride and Et_3N in the presence of imides. The mixture of stereoisomers **7** was prepared in only poor to moderate stereoselectivity (Scheme 5). The ratio of diastereoisomers **7** depends on the substituent on the maleimide skeleton (Table 2).



Scheme 5. Nitrile oxide cycloaddtions to 4a-f.

Compound	R	Z	Ar	Ratio
				endo / exo
4a	CH ₃	Et	Mst	75 : 25
4 b	(CH ₃) ₂ CH	Et	Mst	70:30
4c	(CH ₃) ₂ CHCH ₂	Me	Mst	64 : 36
4c	(CH ₃) ₂ CHCH ₂	Me	Ph	59:41
4d	CH ₂ OH	Et	Mst	59:41
4e	CH ₂ COOEt	Et	Mst	71:29
4 e	CH ₂ COOEt	Et	Ph	72:28
4f	3-Indolylmethyl	Me	Mst	60 : 40

Table 2. Stereoselectivity of 1,3-dipolar cycloaditions of nitrile oxides to maleimides 4a-f.

Conclusion

A modified route for the synthesis of novel chiral maleimides has been developed. The synthetic pathway is short, using only readily available starting materials and reagents. 1,3-Dipolar cycloadditions of nitrile oxides proceeded with moderate diastereoselectivity and further improvement is currently in progress.

Experimental

General

¹H NMR spectra were recorded at 300 MHz on a Varian VXR 300 or at 80 MHz on a Tesla BS 487 at 293 K in CDCl₃. Spectra were internally referenced to TMS. Peaks are reported in ppm downfield of TMS. Multiplicities are reported as singlet (s), doublet (d), triplet (t), quartet (q), some combination of these, broad (br), or multiplet (m). ¹³C NMR spectra were recovered at 75.0 MHz on the same spectrometer as ¹H NMR spectra at 293 K in CDCl₃. Peak assignments were made by DEPT editing of the spectra, and a * indicates that peak assignments may be interchanged. Optical rotations were recorded on POLAR L-µP (IBZ Messtechnik) polarimeter and are reported along with solvent and concentration in g/100ml. Flash chromatography was carried out on 63-200 µm or 40-60 µm silica gel, thin layer chromatography was carried out on aluminium backed silica plates containing UV₂₅₄ by Lachema and plates were visualised with UV light and Mostaine solution as appropriate. New compounds were characterised by their elemental compositions established by combustion analysis. All

yields refer to isolated, spectroscopically pure material, and have not been optimised.

N-Maleonyl-(S)-alanine ethyl ester 4a

Method B

To a solution of *exo*-7-oxohimic acid anhydride (1.00 g, 6.01 mmol) and Et₃N (2.00 ml, 14.0 mmol) dissolved in toluene (50 cm³) was added (*S*)-alanine ethyl ester hydrochloride (1.41g, 7.00mmol) and the resulting mixture was stirred at 110°C for 19 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo, and the residue purified by column chromatography (eluent chloroform-methanol 10:1, 40 g silica gel 40-60 µm mesh, 30x2.5 cm) to give the *title compound* **4a** (330 mg, 28 %) as a pale yellow oil, $\left[\right]_{D}^{D} = -27.3^{\circ}$ (c = 0.11, CH₂Cl₂).

¹H NMR (300 MHz; CDCl₃) 1.25 (3H, t, OCH₂CH₃), 1.62 (3H, d, *J* 7.3 Hz, CH₃), 4.18 (2H, q, O*CH*₂CH₃), 4.75 (1H, q, *J* 7.5 *Hz*, N-C*H*), 6.75 (2H, s, =CH).

N-Maleonyl-(S)-valine-ethyl ester 4b

Method B

To a solution of *exo*-7-oxohimic acid anhydride (750 mg, 4.50 mmol) and Et₃N (1.53 ml, 11.0 mmol) dissolved in toluene (50 cm³) was added (*S*)-valine ethyl ester hydrochloride (1.00 g, 5.50 mmol) and the resulting mixture was stirred at 110°C for 36 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo,

and the residue purified by column chromatography (eluent dichloromethane-methanol 12:1, 40g silica gel 40-60 μ m mesh, 30x2.5cm) to give the *title compound* **4b** (500 mg, 50 %) as a yellow oil, [] $_{\rm D}^{25}$ = - 14.7° (c = 0.15, CH₂Cl₂).

¹H NMR (300 MHz; CDCl₃) 0.87 (3H, d, *J* 6.8 Hz, CH₃), 1.08 (3H, d, *J* 6.7 Hz, CH₃), 1.23 (3H, t, OCH₂CH₃), 2.66 (1H, m, CH), 4.17 (2H, q, OCH₂CH₃), 4.38 (1H, d, *J* 8.1 Hz, N-CH), 6.79 (2H, s, =CH).

¹³C NMR (75 MHz; CDCl₃) 14.07 (q, OCH₂CH₃), 19.37 (q, CH₃), 20.99 (q, CH₃), 28.47 (d, CH), 57.98 (d, N-CH), 61.43 (t, OCH₂CH₃), 134.21 (d, =CH), 168.62 (s, C=O), 170.22 (s, C=O).

N-Maleonyl-(S)-leucine-methyl ester 4c

Method A

To a solution of exo-7-oxohimic acid anhydride (2.00 g, 12.0 mmol) and Et₃N (2.10 cm³, 15.0 mmol) dissolved in toluene (100 cm³) was added (S)-leucine methyl ester hydrochloride (2.15 g, 12.0 mmol) and the resulting mixture was stirred at 110°C for 20 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm^3) , filtered and the filtrate was washed with 2M HCl $(3x20 \text{ cm}^3)$ and the aqueous phase back extracted with ethyl acetate (3x30 cm³). The combined organic layer was dried (MgSO₄) and the solvent was evaporated in vacuo. Purification by column chromatography (eluent chloroform, 40 g silica gel 40-60 µm mesh, 30x2.5 cm) to leave the title compound 4c (640 mg, 24 %) as a colourless oil, which solidified on standing. [] $_{\rm D}^{25} = -13.0^{\circ}$ (c = 0.1, CH₂Cl₂).

Method B

To a solution of *exo*-7-oxohimic acid anhydride (1.00 g, 6.01mmol) and Et₃N (2.00 cm³, 14.0mmol) dissolved in toluene (40 cm³) was added (*S*)-leucine methyl ester hydrochloride (1.27 g, 7.00 mmol) and the resulting mixture was stirred at 110°C for 38 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo, and the residue purified by column chromatography (eluent dichloromethane-methanol 12:1, 60 g silica gel 40-60 μ m mesh, 40x2 cm) to give the *title compounds* **4c** (602 mg, 44 %) as a colourless oil.

Method C

To a solution of maleic anhydride (503 mg, 5.09 mmol) and Et_3N (1.05 cm³, 7.50 mmol) dissolved in toluene (50 cm³) was added (*S*)-leucine methyl ester hydrochloride (1.10 g, 6.00 mmol) and the resulting mixture was stirred at 110°C for 46 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and

filtered. The filtrate was evaporated in vacuo, and the residue purified by column chromatography (eluent chloroform-metanol 12:1, 60 g silica gel 40-60 μ m mesh, 40x2 cm) to give the *title compound* **4c** (570 mg, 49 %) as a colourless oil.

¹H NMR (300 MHz; CDCl₃) 0.90 (3H, d, *J* 6.6 Hz, CH₃), 0.94 (3H, d, *J* 6.6 Hz, CH₃), 1.42 (1H, m, CH), 1.85 (1H, ddd, *J* 14.2, 11.7, 4.7Hz, CH₂), 2.25 (1H, ddd, CH₂), 3.72 (3H, s, OCH₃), 4.75 (1H, dd, *J* 11.7,4.3Hz, N-CH), 6.75 (2H, s, =CH).

¹³C NMR (75 MHz; CDCl₃) 20.90 (q, CH₃), 23.10 (q, CH₃), 25.02 (d, CH), 37.06 (t, CH₂), 50.54 (d, N-*C*H), 52.73 (q, OCH₃), 134.29 (d, =CH), 170.17 (s, C=O).

N-Maleonyl-(S)-serine- ethyl ester 4d

Method A

To a solution of *exo*-7-oxohimic acid anhydride (2.00 g, 12.0 mmol) and Et₃N (2.10 cm³, 15.0 mmol) dissolved in toluene (100 cm³) was added (*S*)-serine ethyl ester hydrochloride (2.04 g, 12.0 mmol) and the resulting mixture was stirred at 110°C for 24 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³), filtered and the filtrate was washed with 2M HCl (3x20 cm³) and the aqueous phase back extracted with ethyl acetate (3x30 cm³). The combined organic layer was dried (MgSO₄) and the solvent was evaporated in vacuo to leave the *title compound* **4d** (404 mg, 16 %) as a colourless oil.

¹H NMR (300 MHz; CDCl₃) 1.25 (3H, t, OCH₂*CH*₃), 3.24 (1H, brs, OH), 4.12 (2H, m, CH₂), 4.22 (2H, d, *J* **7.2**, OCH₂CH₃), 4.82 (1H, t, N-C*H*), 6.81 (2H, s, =CH).

¹³C NMR (300 MHz; CDCl₃) 14.04 (q, OCH₂CH₃), 54.66 (d, N-CH), *60.53 (t, CH₂OH), *62.53 (t,OCH₂CH₃), 134.53 (d, =*C*H), 167.81 (s, C=O), 170.52 (2s, C=O).

N-Maleonyl-(S)-aspartic acid diethyl ester 4e

Method A

To a solution of *exo*-7-oxohimic acid anhydride (1.00 g, 6.01 mmol) and Et₃N (1.70 cm³, 12.0 mmol) dissolved in toluene (100 cm³) was added (*S*)-aspartic acid diethyl ester hydrochloride (1.58 g, 7.00 mmol) and the resulting mixture was stirred at 110°C for 29 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³), filtered and the filtrate was washed with 2M HCl (3x20 cm³) and the aqueous phase back extracted with ethyl acetate (3x30 cm³). The combined organic layer was dried (MgSO₄) and the solvent was evaporated in vacuo. Purification by column chromatography (eluent i-hexaneethyl acetate 3:2, 40 g silica gel 40-60 μ m mesh, 35x1.8

cm) to leave the *title compound* 4e (630 mg, 39 %) as a yellow oil.

Method B

To a solution of *exo*-7-oxohimic acid anhydride (1.00 g, 6.01 mmol) and Et₃N (2.00 cm³, 14.0 mmol) dissolved in toluene (50 cm³) was added (*S*)-aspartic acid diethyl ester hydrochloride (1.58 g, 7.00 mmol) and the resulting mixture was stirred at 110°C for 56 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo and the residue purified by column chromatography (eluent chloroform-methanol 12:1, 50 g silica gel 40-60 µm mesh, 35x2 cm) to give the *title compound* **4e** (1.00g, 62 %) as a yellow oil, $\begin{bmatrix} \\ \\ \end{bmatrix}_{D}^{25} = -15.6^{\circ}$ (c = 0.16, CH₂Cl₂).

Method C

To a solution of maleic anhydride (1.00 g, 10.19 mmol) and Et₃N (2.11 cm³, 15.00 mmol) dissolved in toluene (50 cm³) was added (*S*)-aspartic acid diethyl ester hydrochloride (2.48 g, 11.00 mmol) and the resulting mixture was stirred at 110°C for 44 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo, and the residue purified by column chromatography (eluent chloroform-methanol 12:1, 70 g silica gel 40-60 µm mesh, 80x2 cm) to give the *title compound* **4e** (1.3 g, 55 %) as a yellow oil, []_D²⁵ = -16.0° (c = 0.15, CH₂Cl₂).

¹H NMR (300 MHz; CDCl₃) 1.23 (6H, t,t, 2x OCH₂*CH*₃), 3.03 (1H, dd, *J* 16.7, 9.2 Hz, CH₂), 3.25 (1H, dd, CH₂), 4.16 (4H, q,q, 2x O*CH*₂CH₃), 5.18 (1H, dd, *J* 9.2, 5.8 Hz, N-C*H*), 6.76 (2H, s, =CH).

¹³C NMR (300 MHz; CDCl₃) 14.01 (q, OCH₂CH₃), 14.10 (q, OCH₂CH₃), 33.98 (t, CH₂), 48.34 (d, N-CH), 61.08 (t, OCH₂CH₃), 62.35 (t, OCH₂CH₃), 134.38 (d, =CH), 168.08, 169.61, 169.93 (s, C=O).

N-Maleonyl-(S)-tryptophane-methyl ester 4f

Method B

To a solution of *exo*-7-oxohimic acid anhydride (502 mg, 3.00 mmol) and Et₃N (0.83 cm³, 6.00 mmol) dissolved in toluene (40 cm³) was added (S)-tryptophane methyl ester hydrochloride (890 mg, 3.50 mmol) and the resulting mixture was stirred at 110°C for 24 hours. The solution was cooled to room temperature, diluted with ethyl acetate (20 cm³) and filtered. The filtrate was evaporated in vacuo, and the residue purified by column chromatography (eluent i-hexane-ethyl acetate 1:2, 30 g silica gel 40-60 μ m mesh, 35x2 cm) to give the *title compound* **4f** (453 mg, 50 %) as a yellow oil, which solidified on standing. []²⁵_D = -143.6° (c = 0.11, CH₂Cl₂).

¹H NMR (300 MHz; CDCl₃) 3.62 (2H, dd, J 9.8, 3.8Hz, CH₂), 3.75 (3H, s, OCH₃), 5.04 (1H, dd, J 9.6, 6.7Hz, N-CH), 6.50 (2H, s, =CH), 6.88 (1H, d, J 1.83Hz, =CH-N), 7.05 (2H, m, Harom.), 7.21 (1H,d, J 7.68, Harom.), 7.50 (1H, d, Harom.), 8.23 (1H, s, NH).

¹³C NMR (300 MHz; CDCl₃) 24.59 (t, CH₂), 52.67 (d, N-CH), 52.86 (q, OCH₃),110.59, 111.31, 118.34, 119.48, 122.04, 122.78, 127.05, 136.05 (Carom.), 134.89 (d, =*C*H), 169.64 (s, C=O), 170.00 (s, C=O).

1,3-Dipolar cycloaddition of 2,4,6-trimethylphenylnitrile oxide to maleimide **4a-e**

The nitrile oxide (0.4 mmol) was stirred with maleimide **4a-f** in 5 cm³ benzene at 80°C The reaction was controlled by TLC. After completion of this reaction the solution was evaporated and a sample for NMR was carried out.

1,3-Dipolar cycloaddition of phenylnitrile oxide to maleimide **4c,e**

Triethylamine (0.16 cm³,1.2 mmol) in anhydrous ether (20 cm³) was added during 24 hours to a stirred and cooled (0-5°C, salt-ice bath) solution of maleimide (1.0 mmol) and bezhydroxymoyl chloride (1.2 mmol) in anhydrous ether (25 cm³). The resulting reaction mixture was allowed to stir for 7 days at room temperature. After this time the reaction mixture was filtered, the solution was evaporated to dryness and a sample for NMR was carried out.

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Sample Availability: Available from the authors.