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# Asymmetric Synthesis of the Epimeric (3S)-3-((E)-Hex-1-enyl)-2-methylcyclohexanones 

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#### Abstract

The asymmetric rhodium-catalysed 1,4-addition of alkenylzirconium reagents to 2-cyclohexenone can be useful in the synthesis of 3-alkenyl-2-methylcyclohexanones, provided that formaldehyde is used in trapping the intermediate zirconium enolates. In this manner a four-step sequence leading to the two epimeric 3-hexenyl-2methylcyclohexanones in enantiomeric form was developed.


Keywords: Enantioselective 1,4-addition, zirconium O-enolate trapping, 2-cyclohexenone.

## Introduction

In the context of the development of CD-ring modified structural analogs of calcitriol, the hormonally active metabolite of vitamin $\mathrm{D}_{3}$ [1], we required the epimeric ketones $\mathbf{4 a}$ and $\mathbf{4 b}$ (Scheme 1) in enantiopure form. Both cyclohexanones are being converted to derivatives in which the bicyclic CD-entity is replaced by a single six-membered D-ring and in which the hexenyl substituent serves as a latent aldehyde for the introduction of the A-ring [2]. We describe herein in detail a synthesis of $\mathbf{4 a}$ and $\mathbf{4 b}$ featuring a highly enantioselective 1,4-addition of alkenylzirconocene to 2-cyclohexenone catalysed by the chiral rhodium complex generated from $[\mathrm{Rh}(\mathrm{cod}) \mathrm{Cl}]_{2}$ and $(R)$-BINAP [3], followed by trapping of the O-enolate with formaldehyde.

## Results and Discussion

The synthesis of the two epimeric cyclohexanones $\mathbf{4 a}$ and $\mathbf{4 b}$ rests on the asymmetric $1,4-$ introduction of the hexenyl side chain on 2-cyclohexenone followed by trapping of the resulting metal O-enolate with an electrophilic reagent.

Scheme 1. Synthetic pathway to epimeric (3S)-3-((E)-hex-1-enyl)-2-methylcyclohexanones $\mathbf{4 a}$ and $\mathbf{4 b}$.




$3 a \mathbf{a}$
3ab
4a


1b $\mathrm{R}=\mathrm{H}$
 C

3ba


3bb


4b
(a) $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}, n-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{C}=\mathrm{CH}, \mathrm{THF}, \mathrm{rt}, 45 \mathrm{~min} ;\left[\mathrm{Rh}(\operatorname{cod}) \mathrm{Cl}_{2},(R)-\mathrm{BINAP}, \mathrm{THF}, \mathrm{rt}, 30 \mathrm{~min} ; 2-\right.$ cyclohexenone, rt, 3 h . (b) $\mathrm{CH}_{2} \mathrm{O}$ (from paraformaldehyde), $-78{ }^{\circ} \mathrm{C}$ (89\%). (c) $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$ (a: $96 \%$; b: 99\%). (d) $\mathrm{LiAlH}_{4}, \mathrm{Et}_{2} \mathrm{O}$, reflux, 2 h (a: 95\%; b: 98\%). (e) ( COCl$)_{2}$, DMSO, $\mathrm{Et}_{3} \mathrm{~N},-78^{\circ} \mathrm{C} \rightarrow \mathrm{rt}$ (a: $89 \%$; b: $96 \%$ ).

Among several known enantioselective 1,4-additions of organometallic reagents to $\alpha, \beta$-unsaturated ketones, we obtained the best results with the recently reported rhodium(I)-catalysed addition of alkenylzirconocene chlorides with BINAP as chiral ligand [3,4]. This result also follows Nicolaou's report of the tandem reaction of the rhodium-catalysed asymmetric additions of alkenylzirconium reagents followed by trapping of the zirconium enolate by aldehydes [5]. In our case, the 1,4 -addition was performed on 2-cyclohexenone using ( $E$ )-1-hexenylzirconocene chloride [6], prepared from 1-
hexyne and bis(cyclopentadienyl)zirconium chloride hydride $\left(\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}\right.$ or Schwartz reagent) in the presence of a catalytic amount of the $\mathrm{Rh}(\mathrm{I})$-complex $[\mathrm{Rh}(\operatorname{cod}) \mathrm{Cl}]_{2}$ and $(R)$-BINAP as a chiral ligand. As previously observed by Schwartz, we were unable to directly alkylate the intermediate zirconium $O$-enolate [7]; however, reaction with gaseous formaldehyde at $-78{ }^{\circ} \mathrm{C}$ led, after acid work-up, to a 2.7:1 mixture ( $89 \%$ yield) of $\mathbf{1 a}$ and $\mathbf{1 b}$, respectively [8], with an excellent ee (better than 96\%) [9]. The obtained mixture was readily separated by flash chromatography. The assignment of the transand cis-relationship to the alkyl substituents in the $\mathbf{a}$ - and $\mathbf{b}$-series, respectively, rests on the analysis of NMR spectral data of $\mathbf{1 a}$. The diequatorial orientation in 1a follows from the large vicinal $J$-value of 11.7 Hz for the coupling between $\mathrm{H}_{\mathrm{a}}$ and $\mathrm{H}_{\mathrm{b}}$. The same coupling in epimer $\mathbf{1 b}(5.3 \mathrm{~Hz})$ is indicative of a cis-relationship (Figure 1). The absolute configuration was assigned on the basis of the results obtained by Oi and Inoue [3].

Figure 1. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ structural assignment of $\mathbf{1 a}$.


Figure 2. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ structural assignments of the intermediate alcohols 3aa, 3ab, 3ba, and 3bb.


3aa
$\delta \mathrm{H}_{\mathrm{a}}=3.88 \mathrm{ppm}$
$\Sigma J=25 \mathrm{~Hz}$
W-coupling with $\mathrm{H}_{\mathrm{b}}$


3ba
$\delta \mathrm{H}_{\mathrm{b}}=1.59 \mathrm{ppm}$
$\delta \mathrm{H}_{\mathrm{c}}=2.41 \mathrm{ppm}$
$\mathrm{H}_{\mathrm{a}}: \mathrm{dt}, \mathrm{J}=3.8,7.7 \mathrm{~Hz}$


3ab
$\delta \mathrm{H}_{\mathrm{a}}=3.15 \mathrm{ppm}$
$\Sigma J=39 \mathrm{~Hz}$

$\delta \mathrm{H}_{\mathrm{b}}=1.97 \mathrm{ppm}$
$\delta \mathrm{H}_{\mathrm{c}}=2.11 \mathrm{ppm}$
$H_{a}: d t, J=10.6,4.1 \mathrm{~Hz}$

The further conversion of $\mathbf{1 a}$ and $\mathbf{1 b}$ to $\mathbf{4 a}$ and $\mathbf{4 b}$, respectively, first involves the reduction of the mesylates $\mathbf{2 a}$ and $\mathbf{2 b}$, obtained by treatment with mesyl chloride and triethylamine in dichloromethane, to afford a mixture of the two epimeric alcohols. From 2a there was obtained in $95 \%$ yield a 1:4
mixture of 3aa and 3ab; from $\mathbf{2 b}$ there was obtained in $98 \%$ yield a 1:4 mixture of $\mathbf{3} \mathbf{b a}$ and $\mathbf{3 b b}$, respectively. Again the structural assignment of the four isomeric alcohols rested on ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis (Figure 3). Within the trans-series distinction between 3aa and 3ab readily follows from the coupling constant pattern of the $\mathrm{H}_{\mathrm{a}}$ proton, which indicates an axial hydroxyl group in 3aa (cf. smaller sum of vicinal $J$-values for $\mathrm{H}_{\mathrm{a}}$ ) and an equatorial one in 3ab (cf. larger sum of vicinal $J$-values for $\mathrm{H}_{\mathrm{a}}$ ). Furthermore a characteristic long range $W$ coupling is observed between $H_{a}$ and $H_{b}$ in 3aa. The structural assignment in the cis-series (3ba and 3bb) on the other hand rests on the observed nOe's in both derivatives, in particular between $\mathrm{H}_{\mathrm{a}}$ and the olefinic proton of the hexenyl side-chain in 3ba. Finally, Swern oxidation of both mixtures led to the desired cyclohexanones $\mathbf{4 a}$ and $\mathbf{4 b}$ [10].

## Conclusions

The two epimeric cyclohexanones $\mathbf{4 a}$ and $\mathbf{4 b}$ have been obtained in four steps starting from 2cyclohexenone. The synthesis was based on the very efficient asymmetric 1,4 -introduction of the 1 hexenyl chain using alkenylzirconocene and a chiral rhodium complex.

## Experimental

## General

Dichloromethane was distilled from $\mathrm{CaH}_{2}$. Diethyl ether and tetrahydrofuran (THF) were distilled from benzophenone ketyl. TLC were run on glass plates precoated with silica gel (Merck, 60F-254). Column chromatography was performed on silica gel (Merck, 230-400 mesh). IR spectra ( KBr films) were recorded on a Perkin-Elmer series 1600 FT-IR spectrometer. ${ }^{1} \mathrm{H}$-NMR and ${ }^{13} \mathrm{C}$-NMR spectra were recorded on a Bruker AM-500 spectrometer operating at $500\left({ }^{1} \mathrm{H}\right)$ and $125 \mathrm{MHz}\left({ }^{13} \mathrm{C}\right)$, respectively. Mass spectra (EI) were recorded on a Hewlett-Packard 5898A spectrometer at 70 eV .
(2R,3S)-3-((E)-Hex-1-enyl)-2-(hydroxymethyl)cyclohexanone (1a) and (2S,3S)-3-((E)-hex-1-enyl)-2(hydroxylmethyl)cyclohexanone (1b).

Dry formaldehyde was prepared as follows: paraformaldehyde is predried overnight in vacuo at 60 ${ }^{\circ} \mathrm{C}$ in a three-necked $100-\mathrm{mL}$, round-bottom flask. The flask is equipped with an inlet for $\mathrm{N}_{2}$ (dried over molecular sieves) and is connected via Teflon tubing to the reaction flask. The latter is then equipped with a $\mathrm{CaCl}_{2}$ drying tube. The entire system is evacuated, filled with $\mathrm{N}_{2}$ and the dried paraformaldehyde is depolymerised in a stream of $\mathrm{N}_{2}$ by heating at $180{ }^{\circ} \mathrm{C}$. To a suspension of $\mathrm{Cp}_{2} \mathrm{ZrHCl}(6.20 \mathrm{~g}, 24 \mathrm{mmol})$ in dry THF $(80 \mathrm{~mL})$ under Ar was added 1-hexyne ( $2.76 \mathrm{~mL}, 24 \mathrm{mmol}$ ) and the mixture was stirred at rt for 45 minutes to give a solution of 1-hexenyl-zirconocene chloride. In a two-necked $250-\mathrm{mL}$ flask under $\mathrm{Ar},[\mathrm{Rh}(\mathrm{cod}) \mathrm{Cl}]_{2}(247 \mathrm{mg}, 1 \mathrm{mmol})$ and ( $R$ )-BINAP (749 mg, 1.2 mmol ) were dissolved in dry THF ( 40 mL ) and the solution was stirred at rt for 0.5 h . To the solution of rhodium catalyst, cyclohexenone ( $1.94 \mathrm{~mL}, 20 \mathrm{mmol}$ ) and the solution of 1-hexenylzirconocene chloride were added and the mixture was stirred at rt for 3 h . The reaction mixture was cooled to -78 ${ }^{\circ} \mathrm{C}$ and quenched with gaseous dry formaldehyde (see above). After warming up the mixture to rt, an aqueous saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 3 mL ) was added and the resulting mixture was stirred for $0.5 \mathrm{~h} . t-$ Butylmethyl ether (MTBE; 200 mL ) was added and the precipitate formed was removed by filtration.

After removal of the solvent under reduced pressure, the residue was purified by flash chromatography on silica gel ( $n$-pentane/EtOAc, $83: 17$ to 75:25) to give cyclohexanones $\mathbf{1 a}(2.70 \mathrm{~g})$ and $\mathbf{1 b}(1.04 \mathrm{~g})$ in 89\% total yield.

1a: $R_{f}$ (isooctane/EtOAc, 1:1) 0.38; $[\alpha]_{\mathrm{D}}{ }^{\mathrm{rt}}-10.0$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR v 3493 (br, m), 2929 (vs), 2871 (vs), 1701 (vs), 1459 (m), 1328 (m), 1222 (m), 1085 (m), 1052 (m), 969 (s) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta$ 5.29 ( $1 \mathrm{H}, \mathrm{dt}, J=15.2,6.7 \mathrm{~Hz}$ ), $5.01(1 \mathrm{H}, \mathrm{ddt}, J=15.2,8.8,1.2 \mathrm{~Hz}$ ), $3.85(1 \mathrm{H}, \mathrm{ddd}, J=11.6,8.0,3.2$ $\mathrm{Hz}), 3.73(1 \mathrm{H}, \mathrm{dt}, J=11.6,6.4 \mathrm{~Hz}), 2.85(1 \mathrm{H}, \mathrm{dd}, J=8.0,6.4 \mathrm{~Hz}), 2.15(1 \mathrm{H}$, app. d, $J=13.4 \mathrm{~Hz})$, 2.04 ( 1 H , dddd, $J=11.7,11.5,8.8,3.7 \mathrm{~Hz}$ ), $1.86(3 \mathrm{H}, \mathrm{m}), 1.75(1 \mathrm{H}, \mathrm{tdd}, J=13.4,6.0,1.0 \mathrm{~Hz}), 1.45$ $(2 \mathrm{H}, \mathrm{m}), 1.22(4 \mathrm{H}, \mathrm{m}), 1.19(1 \mathrm{H}, \mathrm{tt}, J=13.5,3.7 \mathrm{~Hz}), 1.11(1 \mathrm{H}, \mathrm{tdd}, J=13.4,11.5,3.7 \mathrm{~Hz}), 0.85(3$ $\mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}) \mathrm{ppm}$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 212.7(\mathrm{C}=\mathrm{O}), 132.4(=\mathrm{CH}), 131.8(=\mathrm{CH}), 60.6\left(\mathrm{CH}_{2}\right), 56.5$ $(\mathrm{CH}), 45.0(\mathrm{CH}), 41.9\left(\mathrm{CH}_{2}\right), 32.6\left(\mathrm{CH}_{2}\right), 32.4\left(\mathrm{CH}_{2}\right), 31.8\left(\mathrm{CH}_{2}\right), 25.7\left(\mathrm{CH}_{2}\right), 22.5\left(\mathrm{CH}_{2}\right), 14.1\left(\mathrm{CH}_{3}\right)$ ppm; MS m/z (\%) 210 ( ${ }^{+}$, 3), 192 (8), 179 (100), 163 (6), 149 (13), 135 (26), 123 (54), 110 (33), 97 (21), 79 (81), 67 (97), 55 (87), 41 (99).

1b: $R_{f}$ (isooctane/EtOAc, 1:1) 0.28; [ $\left.\alpha\right]_{\mathrm{D}}{ }^{\mathrm{rt}}+7.0$ (c 0.9, $\mathrm{CHCl}_{3}$ ); IR v 3408 (br, m), 2956 (vs), 2930 (vs), 2873 (vs), 1707 (vs), 1460 (m), 1379 (m), 1312 (m), 1238 (w), 1138 (m), 1025 (s), 969 (s) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-$ NMR ( $\mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 5.26(1 \mathrm{H}, \mathrm{dt}, J=15.2,6.4 \mathrm{~Hz}), 5.17(1 \mathrm{H}, \mathrm{dd}, J=15.2,8.9 \mathrm{~Hz}), 4.00(1 \mathrm{H}, \mathrm{ddd}, J=$ $11.5,8.8,2.1 \mathrm{~Hz}$ ), $3.34(1 \mathrm{H}, \mathrm{ddd}, J=11.5,8.5,5.3 \mathrm{~Hz}$ ), $2.42(1 \mathrm{H}, \mathrm{dq}, J=8.9,3.6 \mathrm{~Hz}), 2.36(1 \mathrm{H}, \mathrm{dtd}$, $J=8.8,5.3,1.0 \mathrm{~Hz}$ ), $2.17(1 \mathrm{H}$, app. d, $J=13.8 \mathrm{~Hz}), 1.99(1 \mathrm{H}$, app. d, $J=5.3 \mathrm{~Hz}), 1.82(3 \mathrm{H}, \mathrm{m}), 1.58$ $(1 \mathrm{H}, \mathrm{m}), 1.30(3 \mathrm{H}, \mathrm{m}), 1.19(4 \mathrm{H}, \mathrm{m}), 0.83(3 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 212.0(\mathrm{C}=\mathrm{O})$, $133.1(=\mathrm{CH}), 128.4(=\mathrm{CH}), 61.5\left(\mathrm{CH}_{2}\right), 55.8(\mathrm{CH}), 43.6(\mathrm{CH}), 41.7\left(\mathrm{CH}_{2}\right), 32.4\left(\mathrm{CH}_{2}\right), 31.8\left(\mathrm{CH}_{2}\right)$, $31.7\left(\mathrm{CH}_{2}\right), 22.9\left(\mathrm{CH}_{2}\right), 22.4\left(\mathrm{CH}_{2}\right), 14.0\left(\mathrm{CH}_{3}\right) \mathrm{ppm} ; \mathrm{MS} \mathrm{m} / \mathrm{z}(\%) 210\left(\mathrm{M}^{+}, 2\right), 192(10), 179(14), 163$ (6), 149 (9), 135 (32), 123 (50), 110 (98), 107 (29), 97 (47), 86 (48), 79 (83), 67 (100), 55 (82), 41 (98).
((1R,2S)-2-((E)-Hex-1-enyl)-6-oxocyclohexyl)methyl methanesulfonate (2a) and ((1S,2S)-2-((E)-hex-1-enyl)-6-oxocyclohexyl)methyl methanesulfonate (2b)

To an ice-cold solution of alcohol 1a ( $458 \mathrm{mg}, 2.18 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}$ ( $0.61 \mathrm{~mL}, 4.36 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(22 \mathrm{~mL})$ was added $\mathrm{MsCl}(0.25 \mathrm{~mL}, 3.27 \mathrm{mmol})$. After stirring for 0.5 h , the reaction mixture was quenched with an aqueous saturated $\mathrm{NaHCO}_{3}$ solution and the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic extracts were consecutively washed with a HCl solution ( $0.5 \mathrm{M} ; 5 \mathrm{~mL}$ ) and a saturated $\mathrm{NaHCO}_{3}$ solution ( 8 mL ), dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure to give mesylate 2a ( $604 \mathrm{mg}, 96 \%$ ). The same procedure applied to alcohol $\mathbf{1 b}$ afforded mesylate $\mathbf{2 b}$ as a solid in $99 \%$ yield.

2a: $R_{f}$ (isooctane/EtOAc, 6:4) 0.37; $[\alpha]_{\mathrm{D}}{ }^{\mathrm{rt}}-17.0\left(c 1.0, \mathrm{CH}_{3} \mathrm{OH}\right.$ ); IR $\vee 2956$ (s), 2931 (s), $2860(\mathrm{~m})$, 1715 (vs), 1458 (m), 1356 (vs), 1251 (vw), 1174 (vs), 974 (s), 950 (s), 823 (w), 529 (s) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 5.37(1 \mathrm{H}, \mathrm{dt}, J=15.3,6.6 \mathrm{~Hz}), 4.95(1 \mathrm{H}, \mathrm{dd}, J=15.3,9.0 \mathrm{~Hz}), 4.32(1 \mathrm{H}, \mathrm{dd}, J=9.5,1.9$ $\mathrm{Hz}), 4.21(1 \mathrm{H}, \mathrm{dd}, J=9.5,5.7 \mathrm{~Hz}), 2.49(3 \mathrm{H}, \mathrm{s}), 2.15(1 \mathrm{H}, \mathrm{app} . \mathrm{d}, J=13.9 \mathrm{~Hz}), 2.02(1 \mathrm{H}, \mathrm{tdd}, J=$ $11.8,9.0,3.7 \mathrm{~Hz}$ ), $1.90(2 \mathrm{H}, \mathrm{m}), 1.75(1 \mathrm{H}, \mathrm{ddd}, \mathrm{J}=11.8,5.7,1.9), 1.69(1 \mathrm{H}, \mathrm{td}, J=13.9,6.0 \mathrm{~Hz})$, $1.42(2 \mathrm{H}, \mathrm{m}), 1.27(4 \mathrm{H}, \mathrm{m}), 1.15(1 \mathrm{H}, \mathrm{qt}, J=14.2,3.5 \mathrm{~Hz}), 1.02(1 \mathrm{H}, \mathrm{qd}, J=12.9,3.3 \mathrm{~Hz}), 0.91(3$
$\mathrm{H}, \mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 206.5(\mathrm{C}=\mathrm{O}), 132.9(=\mathrm{CH}), 131.0(=\mathrm{CH}), 66.0\left(\mathrm{OCH}_{2}\right), 53.4$ $(\mathrm{CH}), 44.4(\mathrm{CH}), 41.2\left(\mathrm{CH}_{2}\right), 36.3\left(\mathrm{SCH}_{3}\right), 32.4\left(\mathrm{CH}_{2}\right), 32.2\left(\mathrm{CH}_{2}\right), 31.6\left(\mathrm{CH}_{2}\right), 25.1\left(\mathrm{CH}_{2}\right), 22.4$ $\left(\mathrm{CH}_{2}\right), 14.0\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$; MS m/z (\%) $289\left(\mathrm{MH}^{+}, 8\right), 205$ (4), 192 (79), 179 (17), 163 (34), 149 (36), 135 (65), 122 (66), 107 (36), 93 (43), 79 (100), 67 (42), 55 (46), 41 (47).

2b: mp $39{ }^{\circ} \mathrm{C}$; $R_{f}$ (isooctane/EtOAc, 6:4) 0.37; $[\alpha]_{\mathrm{D}}{ }^{\mathrm{rt}}-28.5$ (c 1.0, CH3OH); IR v 2956 (s), 2932 (s), 2873 (m), 1711 (vs), 1459 (w), 1358 (vs), 1210 (vw), 1177 (vs), 1142 (vw), 959 (vs), 866 (w), 825 (w), $528(\mathrm{~m}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 5.32(1 \mathrm{H}, \mathrm{dt}, J=-15.1,6.8 \mathrm{~Hz}), 5.07(1 \mathrm{H}, \mathrm{dd}, J=15.1,9.7$ $\mathrm{Hz}), 4.51(1 \mathrm{H}, \mathrm{dd}, J=10.2,6.5 \mathrm{~Hz}), 4.00(1 \mathrm{H}, \mathrm{dd}, J=10.2,6.8 \mathrm{~Hz}), 2.61(1 \mathrm{H}, \mathrm{dq}, J=9.7,3.4 \mathrm{~Hz})$, $2.48(1 \mathrm{H}$, app. q, $J=6.2 \mathrm{~Hz}), 2.26(3 \mathrm{H}, \mathrm{s}), 2.11(1 \mathrm{H}$, app. d, $J=13.2 \mathrm{~Hz}), 1.80(2 \mathrm{H}, \mathrm{m}), 1.73(1 \mathrm{H}$, td, $J=13.4,6.0 \mathrm{~Hz}), 1.51(1 \mathrm{H}, \mathrm{m}), 1.35(3 \mathrm{H}, \mathrm{m}), 1.18(4 \mathrm{H}, \mathrm{m}), 0.84(3 \mathrm{H}, \mathrm{t}, J=7.0 \mathrm{~Hz}) \mathrm{ppm}$; ${ }^{13} \mathrm{C}-$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 208.0(\mathrm{C}=\mathrm{O}), 134.9(=\mathrm{CH}), 126.3(=\mathrm{CH}), 68.0\left(\mathrm{OCH}_{2}\right), 52.4(\mathrm{CH}), 43.4(\mathrm{CH}), 41.5$ $\left(\mathrm{CH}_{2}\right), 36.2\left(\mathrm{SCH}_{3}\right), 32.3\left(\mathrm{CH}_{2}\right), 31.5\left(\mathrm{CH}_{2}\right)$, $31.2\left(\mathrm{CH}_{2}\right), 22.9\left(\mathrm{CH}_{2}\right), 22.3\left(\mathrm{CH}_{2}\right), 13.9\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$; MS m/z (\%) 256 (5), 223 (4), 208 (14), 192 (24), 149 (24), 135 (16), 123 (14), 105 (16), 91 (35), 79 (56), 67 (51), 55 (76), 41 (100).
(2S,3S)-3-((E)-Hex-1-enyl)-2-methylcyclohexanol (3aa, 3ab) and (2R,3S)-3-((E)-hex-1-enyl)-2-methylcyclohexanol (3ba, 3bb)
$\mathrm{LiAlH}_{4}(143 \mathrm{mg}, 3.76 \mathrm{mmol})$ in dry $\mathrm{Et}_{2} \mathrm{O}(7.5 \mathrm{~mL})$ was refluxed for 30 min . Methanesulfonate 2a ( $546 \mathrm{mg}, 1.89 \mathrm{mmol}$ ) in dry $\mathrm{Et}_{2} \mathrm{O}(7.5 \mathrm{~mL}$ ) was dropwise added at reflux temperature over a period of 10 min . The reaction mixture was refluxed for 2 h , then cooled to $0^{\circ} \mathrm{C}$ and quenched by the sequential addition of $\mathrm{H}_{2} \mathrm{O}(145 \mu \mathrm{~L}), \mathrm{NaOH}(15 \%$ solution; $145 \mu \mathrm{~L})$ and $\mathrm{H}_{2} \mathrm{O}(300 \mu \mathrm{~L})$. After stirring at rt for 1 h , the white precipitate was removed by filtration over Celite ${ }^{\circledR}$ and washed with EtOAc. The filtrate was concentrated under reduced pressure and the residue was purified by flash chromatography on silica gel ( $n$-pentane/EtOAc, 9:1) to give alcohols 3aa ( 70 mg ) and 3ab ( 285 mg ) in $96 \%$ total yield. The same procedure applied to methanesulfonate $\mathbf{2 b}(1.06 \mathrm{~g})$ afforded alcohols $\mathbf{3 b a}(141 \mathrm{mg})$ and $\mathbf{3 b b}$ ( 563 mg ) in $98 \%$ total yield.

3aa: $R_{f}$ (isooctane/EtOAc, 6:4) 0.54; [ $\left.\alpha\right]_{\mathrm{D}}{ }^{\mathrm{rt}}-45.7$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR v 3392 (br, m), 2957 (s), 2928 (vs), 2872 (s), 1456 (m), 1376 (w), 1212 (w), 963 (w), 879 (w) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 5.37$ ( $1 \mathrm{H}, \mathrm{dt}$, $J=15.2,6.2 \mathrm{~Hz}$ ), 5.15 ( $1 \mathrm{H}, \mathrm{ddt}, J=15.2,8.6,1.1$ ), $3.88(1 \mathrm{H}, \mathrm{m}, \Sigma \mathrm{J}=25 \mathrm{~Hz}), 1.96(3 \mathrm{H}, \mathrm{m}), 1.81$ ( 1 H, app. d, $J=13.2 \mathrm{~Hz}$ ), $1.64(2 \mathrm{H}, \mathrm{m}), 1.46(2 \mathrm{H}, \mathrm{m}), 1.35-1.24(5 \mathrm{H}, \mathrm{m}), 1.11(1 \mathrm{H}, \mathrm{m}), 0.92(3 \mathrm{H}, \mathrm{d}, J$ $=7.0 \mathrm{~Hz}), 0.87(3 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}) \mathrm{ppm}$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 134.7(=\mathrm{CH}), 130.1(=\mathrm{CH}), 71.1(\mathrm{OCH})$, $41.7(\mathrm{CH}), 40.6(\mathrm{CH}), 33.3\left(\mathrm{CH}_{2}\right), 33.3\left(\mathrm{CH}_{2}\right), 32.3\left(\mathrm{CH}_{2}\right), 31.9\left(\mathrm{CH}_{2}\right), 22.2\left(\mathrm{CH}_{2}\right), 19.6\left(\mathrm{CH}_{2}\right), 16.7$ $\left(\mathrm{CH}_{3}\right), 14.0\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$; MS m/z (\%) 196 ( ${ }^{+}$, <1), 178 (54), 163 (8), 149 (25), 135 (56), 121 (25), 111 (50), 93 (51), 79 (52), 67 (54), 55 (54), 41 (100).

3ab: $R_{f}$ (isooctane/EtOAc, 6:4) 0.47; $[\alpha]_{\mathrm{D}}{ }^{\mathrm{rt}}+12.3$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR v 3350 (br, m), 2958 (s), 2926 (vs), 2857 (s), 1458 (m), 1376 (w), 1358 (w), 1294 (vw), 1117 (vw), 1032 (s), 966 (vs), 858 ( vw$)_{\mathrm{cm}^{-1} \text {; }}$ ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 5.37(1 \mathrm{H}, \mathrm{dt}, J=15.2,6.7 \mathrm{~Hz}), 5.17(1 \mathrm{H}, \mathrm{ddd}, J=15.2,8.5,1.2), 3.15(1 \mathrm{H}, \mathrm{m}$, $\Sigma J=39 \mathrm{~Hz}), 1.96(3 \mathrm{H}, \mathrm{m}), 1.72(1 \mathrm{H}$, app. d, $J=13.0 \mathrm{~Hz}), 1.57(2 \mathrm{H}, \mathrm{m}), 1.43(1 \mathrm{H}, \mathrm{s}), 1.37-1.20(6$ $\mathrm{H}, \mathrm{m}), 1.10(1 \mathrm{H}, \mathrm{m}), 1.06(1 \mathrm{H}, \mathrm{m}), 0.97(3 \mathrm{H}, \mathrm{d}, J=6.6 \mathrm{~Hz}), 0.87(3 \mathrm{H}, \mathrm{t}, J=6.9 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-\mathrm{NMR}$
$\left(\mathrm{CDCl}_{3}\right) \delta 134.0(=\mathrm{CH}), 130.4(=\mathrm{CH}), 76.1(\mathrm{OCH}), 47.3(\mathrm{CH}), 44.6(\mathrm{CH}), 35.5\left(\mathrm{CH}_{2}\right), 33.4\left(\mathrm{CH}_{2}\right)$, $32.2\left(\mathrm{CH}_{2}\right), 31.8\left(\mathrm{CH}_{2}\right), 23.9\left(\mathrm{CH}_{2}\right)$, $22.2\left(\mathrm{CH}_{2}\right), 16.0\left(\mathrm{CH}_{3}\right), 14.0\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$; MS m/z (\%) $196\left(\mathrm{M}^{+}\right.$, <1), 178 (22), 163 (5), 149 (22), 135 (29), 121 (42), 108 (21), 93 (40), 79 (47), 67 (52), 55 (59), 41 (100).

3ba: $R_{f}$ (isooctane/EtOAc, 8:2) 0.47; $[\alpha]_{\mathrm{D}}{ }^{\text {rt }}-40.3$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR $v 3349$ (br, m), 2956 (s), 2928 (vs), 2872 (s), 1460 (m), 1377 (w), 1142 (w), 1042 (m), 1017 (m), 967 (s) cm ${ }^{-1} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ $5.40(2 \mathrm{H}, \mathrm{m}), 3.56(1 \mathrm{H}, \mathrm{td}, J=7.7,3.8 \mathrm{~Hz}), 2.41(1 \mathrm{H}, \mathrm{m}), 2.20(2 \mathrm{H}, \mathrm{m}), 1.85(1 \mathrm{H}, \mathrm{m}), 1.59(3 \mathrm{H}$, m), $1.50(2 \mathrm{H}, \mathrm{m}), 1.41-1.28(5 \mathrm{H}, \mathrm{m}), 0.90(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 0.89(3 \mathrm{H}, \mathrm{t}, J=6.5 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 131.2(=\mathrm{CH}), 130.9(=\mathrm{CH}), 72.4(\mathrm{CH}), 41.8(\mathrm{CH}), 41.6(\mathrm{CH}), 32.7\left(\mathrm{CH}_{2}\right), 32.5\left(\mathrm{CH}_{2}\right)$, $31.9\left(\mathrm{CH}_{2}\right), 29.6\left(\mathrm{CH}_{2}\right), 22.2\left(\mathrm{CH}_{2}\right)$, $20.4\left(\mathrm{CH}_{2}\right), 14.5\left(\mathrm{CH}_{3}\right), 14.0\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$, MS m/z (\%) $196\left(\mathrm{M}^{+}\right.$, 4), 178 (14), 163 (4), 149 (17), 135 (22), 121 (42), 111 (39), 93 (42), 79 (48), 67 (47), 55 (49), 41 (100).

3bb: $R_{f}$ (isooctane/EtOAc, 8:2) 0.47; $[\alpha]_{\mathrm{D}}{ }^{\mathrm{rt}}-10.3$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR $v 3350$ (br, m), 2928 (vs), 2860 (s), 1466 (m), 1447 (m), 1378 (w), 1342 (w), 1301 (w), 1119 (w), 1053 (m), 1016 (m), 968 (s) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 5.41(1 \mathrm{H}, \mathrm{dd}, J=15.5,5.3 \mathrm{~Hz}), 5.38(1 \mathrm{H}, \mathrm{dt}, J=15.5,5.8 \mathrm{~Hz}), 3.72(1 \mathrm{H}, \mathrm{dt}, J=$ $10.6,4.1 \mathrm{~Hz}), 2.11(1 \mathrm{H}, \mathrm{m}), 1.97(3 \mathrm{H}, \mathrm{m}), 1.72(1 \mathrm{H}, \mathrm{m}), 1.59(1 \mathrm{H}, \mathrm{app} . \mathrm{d}, \mathrm{J}=12.3 \mathrm{~Hz}), 1.42(1 \mathrm{H}$, m); 1.37-1.20 ( $7 \mathrm{H}, \mathrm{m}$ ), $0.87(3 \mathrm{H}, \mathrm{t}, J=6.8 \mathrm{~Hz}), 0.80(3 \mathrm{H}, \mathrm{d}, J=7.0 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ $133.3(=\mathrm{CH}), 129.5(=\mathrm{CH}), 73.6(\mathrm{CH}), 42.8(\mathrm{CH}), 39.8(\mathrm{CH}), 32.4\left(\mathrm{CH}_{2}\right), 31.9\left(\mathrm{CH}_{2}\right), 29.0\left(\mathrm{CH}_{2}\right), 24.8$ $\left(\mathrm{CH}_{2}\right), 23.2\left(\mathrm{CH}_{2}\right), 22.2\left(\mathrm{CH}_{2}\right), 14.0\left(\mathrm{CH}_{3}\right), 6.7\left(\mathrm{CH}_{3}\right) \mathrm{ppm} ; \mathrm{MS} \mathrm{m} / \mathrm{z}(\%) 196\left(\mathrm{M}^{+},<1\right), 178(40), 163$ (6), 149 (22), 135 (32), 122 (28), 111 (36), 108 (25), 93 (36), 79 (50), 67 (34), 55 (51), 41 (100).
(2S,3S)-3-((E)-Hex-1-enyl)-2-methylcyclohexanone (4a) and (2R,3S)-3-((E)-hex-1-enyl)-2-methylcyclohexanone (4b)

To a solution of $(\mathrm{COCl})_{2}(113 \mu \mathrm{~L}, 1.33 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ was dropwise added a solution of dimethyl sulfoxide (DMSO; $189 \mu \mathrm{~L}, 2.66 \mathrm{mmol}$ ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(750 \mu \mathrm{~L})$ at $-78{ }^{\circ} \mathrm{C}$. After 2 minutes of stirring, a mixture of alcohols 3aa and 3ab ( $238 \mathrm{mg}, 1.21 \mathrm{mmol}$ ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$ was dropwise added over a period of 5 min and the mixture was stirred for $15 \mathrm{~min} . \mathrm{Et}_{3} \mathrm{~N}(845 \mu \mathrm{~L}, 6.66$ mmol ) was then added and the reaction mixture was stirred for an additional 5 min , followed by slow warming to rt . $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ was added and the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were concentrated under reduced pressure and the residue was diluted with $t$-butyl methyl ether (MTBE). After washing with $\mathrm{H}_{2} \mathrm{O}$, the organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$ and evaporated under reduced pressure. The residue was purified by flash chromatography on silica gel ( $n$-pentane/EtOAc, 98:2) to give cyclohexanone $\mathbf{4 a}$ ( $209 \mathrm{mg}, 89 \%$ ). The same procedure applied to the mixture of alcohols 3ba and $\mathbf{3 b b}$ afforded cyclohexanone $\mathbf{4 b}$ in $96 \%$ yield.

4a: $R_{f}$ (isooctane/EtOAc, 8:2) 0.44; [ $\left.\alpha\right]_{\mathrm{D}}{ }^{\mathrm{rt}}-12.0$ (c 1.0, $\mathrm{CH}_{3} \mathrm{OH}$ ); IR v 2930 (vs), 2863 (vs), 1712 (vs), 1450 (m), 1376 (w), 1329 (w), 1307 (w), 1216 (w), 1181 (w), 1128 (bw), 1017 (w), 969 (s) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 5.24(1 \mathrm{H}, \mathrm{dt}, J=15.2,6.6 \mathrm{~Hz}), 5.05(1 \mathrm{H}, \mathrm{dd}, J=15.2,8.4 \mathrm{~Hz}), 2.26(1 \mathrm{H}$, app. d, $J=$ 13.4 Hz ), $1.92(2 \mathrm{H}, \mathrm{m}), 1.86(1 \mathrm{H}, \mathrm{td}, J=13.6,6.0 \mathrm{~Hz}), 1.77(1 \mathrm{H}, \mathrm{m}), 1.73(1 \mathrm{H}, \mathrm{m}), 1.56(1 \mathrm{H}, \mathrm{m})$, $1.48(1 \mathrm{H}, \mathrm{app} . \mathrm{d}, J=13.0 \mathrm{~Hz}), 1.32(1 \mathrm{H}, \mathrm{m}), 1.26(4 \mathrm{H}, \mathrm{m}), 1.17(1 \mathrm{H}, \mathrm{m}), 1.11(3 \mathrm{H}, \mathrm{d}, J=6.0 \mathrm{~Hz})$,
$0.86(3 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}) \mathrm{ppm}$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 209.5(\mathrm{C}=\mathrm{O}), 133.4(=\mathrm{CH}), 130.9(=\mathrm{CH}), 50.4(\mathrm{CH})$, $49.4(\mathrm{CH}), 41.5\left(\mathrm{CH}_{2}\right), 32.9\left(\mathrm{CH}_{2}\right), 32.3\left(\mathrm{CH}_{2}\right), 31.1\left(\mathrm{CH}_{2}\right), 26.0\left(\mathrm{CH}_{2}\right), 21.6\left(\mathrm{CH}_{2}\right), 14.1\left(\mathrm{CH}_{3}\right), 13.0$ $\left(\mathrm{CH}_{3}\right) \mathrm{ppm} ; \mathrm{MS} \mathrm{m} / \mathrm{z}(\%) 194$ ( $\mathrm{M}^{+}, 9$ ), 137 (4), 123 (30), 110 (19), 96 (4), 81 (45), 79 (32), 67 (100), 47 (67).

4b: $R_{f}$ (isooctane/EtOAc, 8:2) 0.44; [ $\left.\alpha\right]_{\mathrm{D}}{ }^{\mathrm{rt}}-13.1$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR $v 2957$ (s), 2930 (vs), 2872 ( s ), 1713 (vs), 1448 (m), 1377 (vw), 1312 (vw), 1222 (w), 1141 (w), 1078 (w), 995 (w), 968 (m) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}-$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 5.32(1 \mathrm{H}, \mathrm{dtd}, J=15.3,6.8,1.1 \mathrm{~Hz}), 5.20(1 \mathrm{H}, \mathrm{ddd}, J=15.3,8.8,0.9 \mathrm{~Hz}), 2.36(1 \mathrm{H}$, $\mathrm{m}), 2.20(2 \mathrm{H}, \mathrm{m}), 1.87(3 \mathrm{H}, \mathrm{m}), 1.65(1 \mathrm{H}, \mathrm{m}), 1.44(3 \mathrm{H}, \mathrm{m}), 1.22(4 \mathrm{H}, \mathrm{m}), 0.97(3 \mathrm{H}, \mathrm{dd}, J=6.8$, $1.3 \mathrm{~Hz}), 0.83(3 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}) \mathrm{ppm} ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 210.6(\mathrm{C}=\mathrm{O}), 133.1(=\mathrm{CH}), 129.0(=\mathrm{CH})$, $48.5(\mathrm{CH}), 47.0(\mathrm{CH}), 41.0\left(\mathrm{CH}_{2}\right), 32.6\left(\mathrm{CH}_{2}\right), 32.0\left(\mathrm{CH}_{2}\right), 31.1\left(\mathrm{CH}_{2}\right), 23.5\left(\mathrm{CH}_{2}\right), 22.5\left(\mathrm{CH}_{2}\right), 14.1$ $\left(\mathrm{CH}_{3}\right), 12.9\left(\mathrm{CH}_{3}\right) \mathrm{ppm}$; MS m/z (\%) $194\left(\mathrm{M}^{+}, 8\right), 179$ (2), 149 (5), 138 (3), 123 (34), 110 (22), 95 (14), 81 (57), 67 (67), 55 (49), 49 (100), 41 (93).

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## References and Notes

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10. The ee of $\mathbf{4 a}$ and $\mathbf{4} \mathbf{b}$ was determined by HPLC based on the racemate using a chiral stationary phase column (Daicel CHIRALPAK® AD-H, eluent: $n$-hexane/EtOH, 99:1); for 4a: $96.3 \%$ ee, for 4b: better than $91 \%$ ee.

Sample Availability: No samples available.
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